



ALMA MATER STUDIORUM  
UNIVERSITÀ DI BOLOGNA

[www.tribchem.it](http://www.tribchem.it)

# Materials Modeling and Design: Tribology and Hydrogen production as case studies

**Maria Clelia Righi**

Department of Physics and Astronomy Augusto Righi

- **Designing Materials by High Throughput Calculations**
- **Modeling Materials Function by Molecular Dynamics**
- **What is tribology ?**
- **High throughput calculations to design solid interfaces**
- **Molecular dynamics to unravel mechanism of function of lubricants**
- **Simulating what happens in a reactor: H<sub>2</sub> and CNT production from CH<sub>4</sub>**



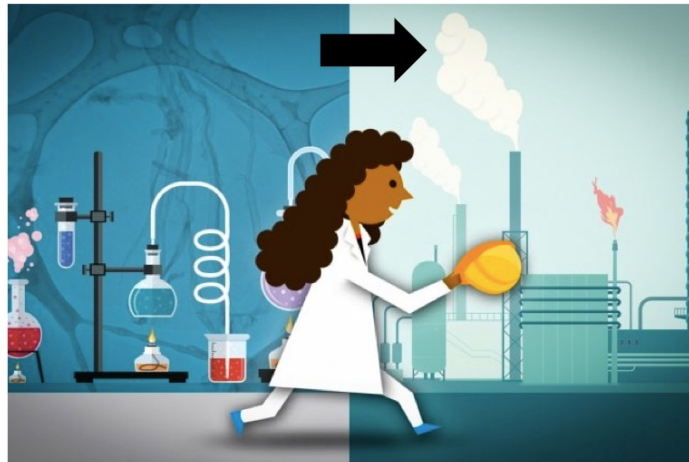
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# Advancing technologies by materials innovation

The advancement of many technologies important for our everyday life requires **materials innovation**.

Developing new materials is a difficult and time consuming task.



**Identify new materials earlier on** to begin the scale-up as soon as possible

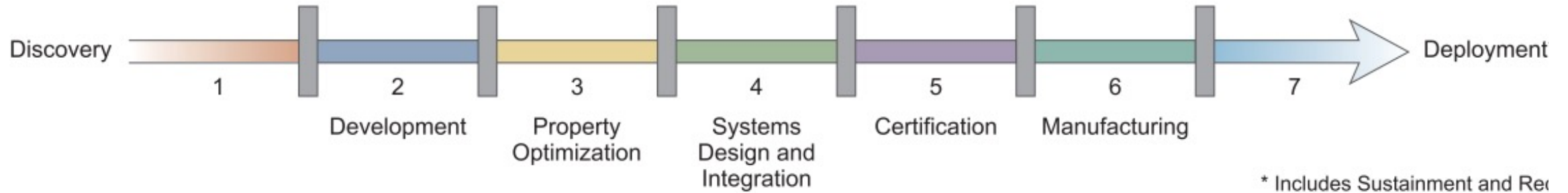
**Identify the strength and weakness of materials during the design process**





# Computational tools and digital data for accelerating materials innovation

**20 years** from the laboratory to the widespread adoption



**Reducing the time and the costs** of material discovery and deployment by **computational tools and digital data**



# Materials design by high throughput calculations

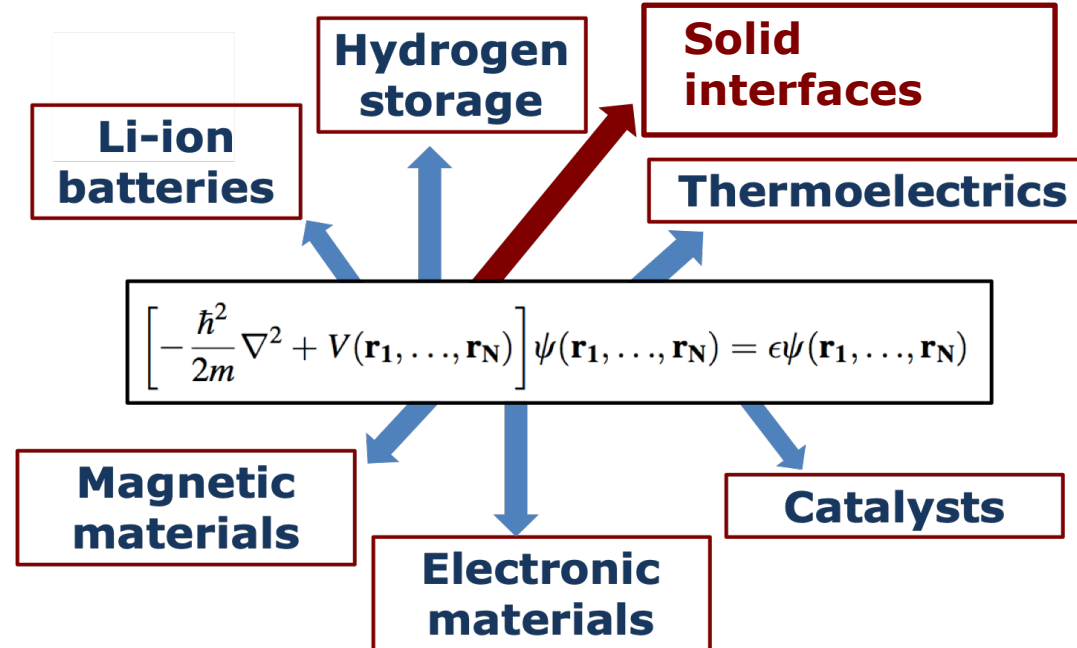
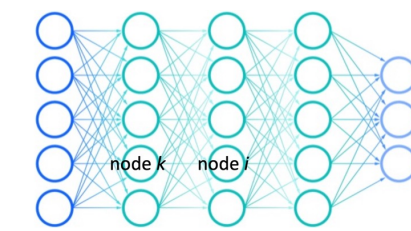
*Ab initio* calculations have become ubiquitous in material science

- ✓ availability of robust computer programs
- ✓ increase of high performance computing (HPC)
- ✓ appearance of curated materials databases



Adhesion Energy (J/m<sup>2</sup>)

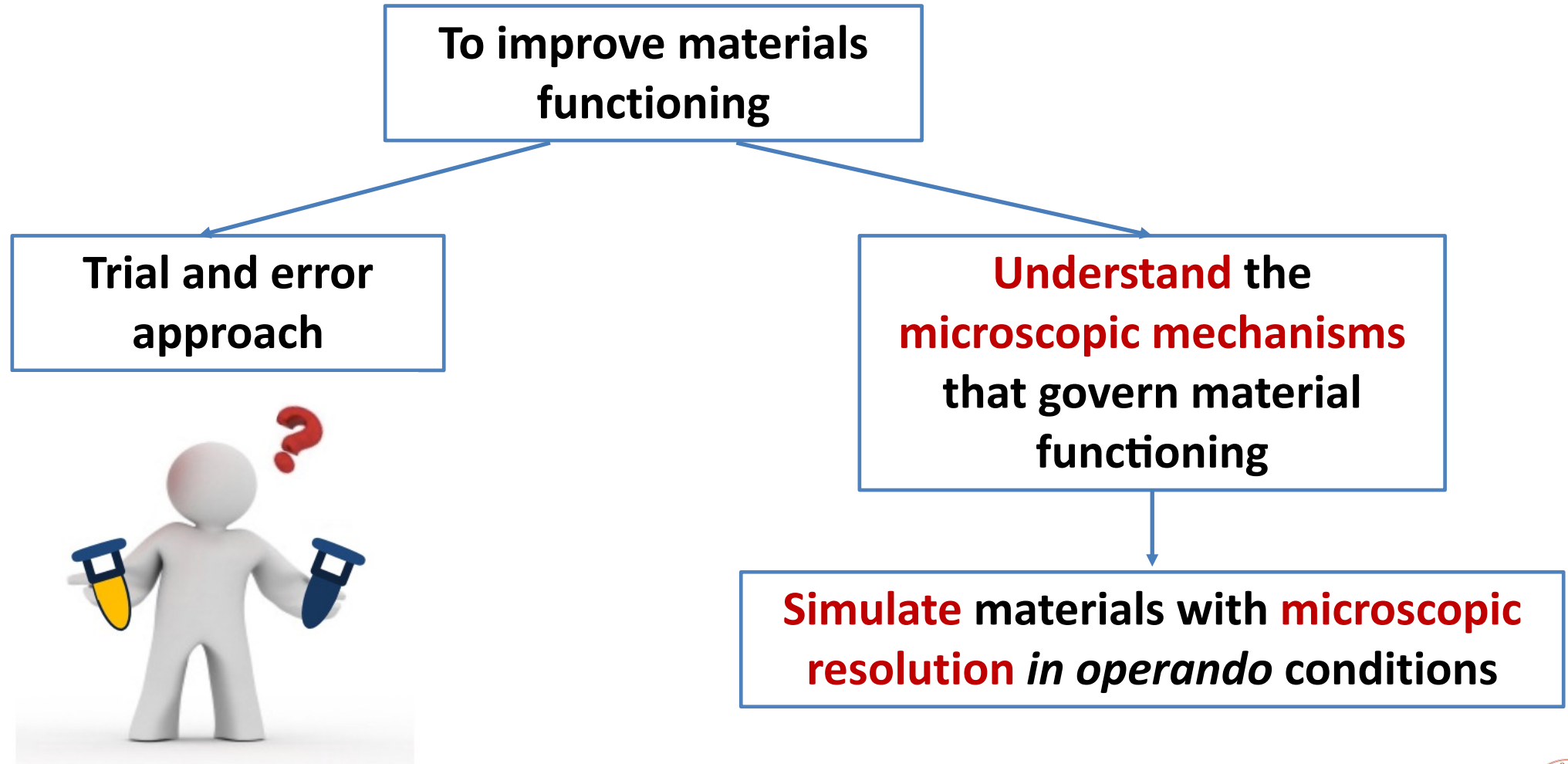
W(110)	-5.87	-4.54	-5.50	-4.29	-5.38	-4.51	-4.49	-4.70	-3.44	-3.96	-2.15	-2.66	-2.39	-2.38
Cr(110)	-4.54	-5.76	-4.46	-5.24	-5.27	-3.27	-4.10	-5.22	-3.19	-3.85	-2.12	-2.54	-2.23	-2.32
Mo(110)	-5.90	-4.46	-5.12	-3.85	-5.26	-3.95	-4.06	-4.20	-3.31	-3.90	-2.14	-2.51	-2.36	-2.33
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Ir(111)	-5.38	-5.27	-5.26	-4.21	-4.41	-3.65	-5.16	-3.41	-2.99	-3.66	-2.02	-2.08	-2.71	-2.37
Ni(111)	-4.51	-3.27	-3.95	-4.12	-3.65	-3.74	-4.39	-3.29	-2.75	-3.32	-2.05	-2.17	-2.39	-2.17
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Au(111)	-2.66	-2.54	-2.51	-2.61	-2.08	-2.17	-3.47	-1.84	-1.80	-1.79	-1.56	-1.33	-2.00	-1.24
Mg(001)	-2.39	-2.23	-2.36	-2.10	-2.71	-2.39	-1.99	-3.18	-1.86	-1.19	-1.41	-2.00	-1.21	-0.97
Zn(001)	-2.38	-2.32	-2.33	-2.49	-2.37	-2.17	-2.22	-2.27	-1.53	-0.81	-1.08	-1.24	-0.97	-0.89



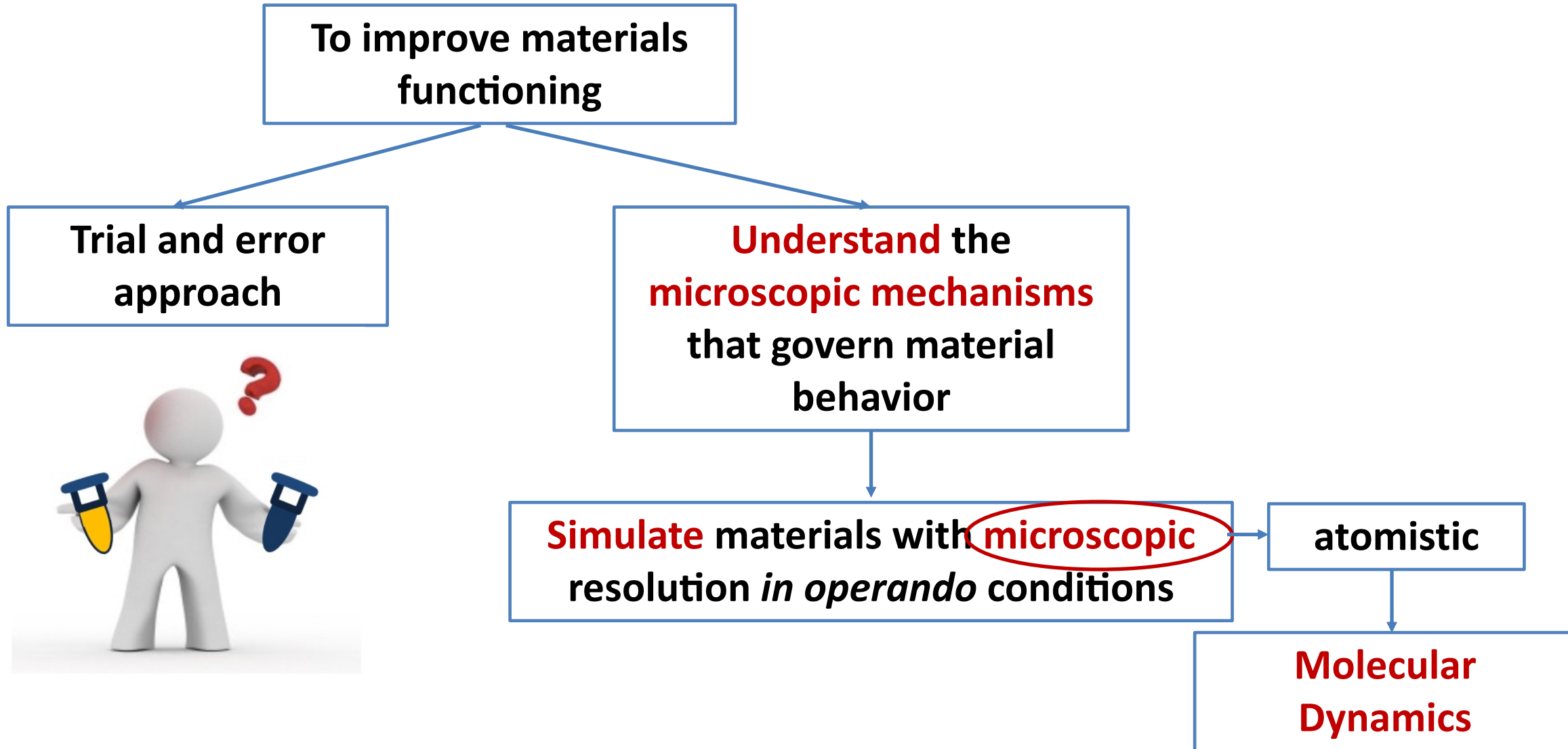
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# Modeling materials function



# Modeling materials function



# Molecular Dynamics (MD)

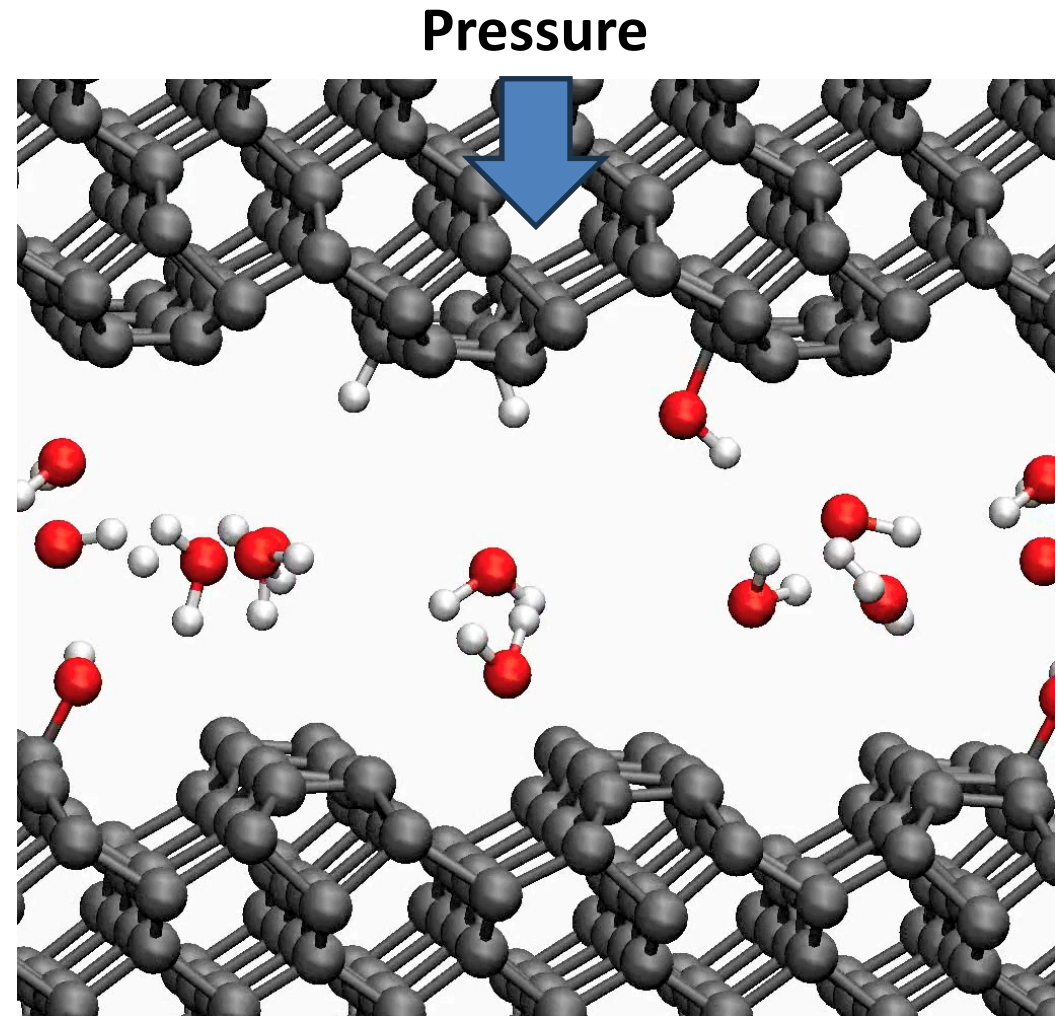
Impose the working conditions,  
Temperature, Pressure, External forces..

Solve the equation of motion for each  
atom in the system

$$\mathbf{a} = \mathbf{F}/m$$

$$\mathbf{F} = -\nabla E(\mathbf{R})$$

$E(\mathbf{R}_1, \mathbf{R}_2, \dots, \mathbf{R}_N)$  interaction energy of the  
atoms



$$E(R_1, R_2, \dots, R_N)$$

### Ab initio MD

$$\mathbf{H} \Psi = \mathbf{E} \Psi$$

solving the **quantum mechanical** equations for the system **electrons**

😊 High accuracy

☹ High computational cost

### Classical MD

$$E = \sum_{ij} v(r_{ij})$$

E sum of **parametric potentials** that **mimic the effects of electrons**

😊 High simulation efficiency

☹ Questionable accuracy

### Machine Learning MD

E is the output layer of a **neural network** trained with *ab initio* data

😊 High accuracy

😊 High simulation efficiency



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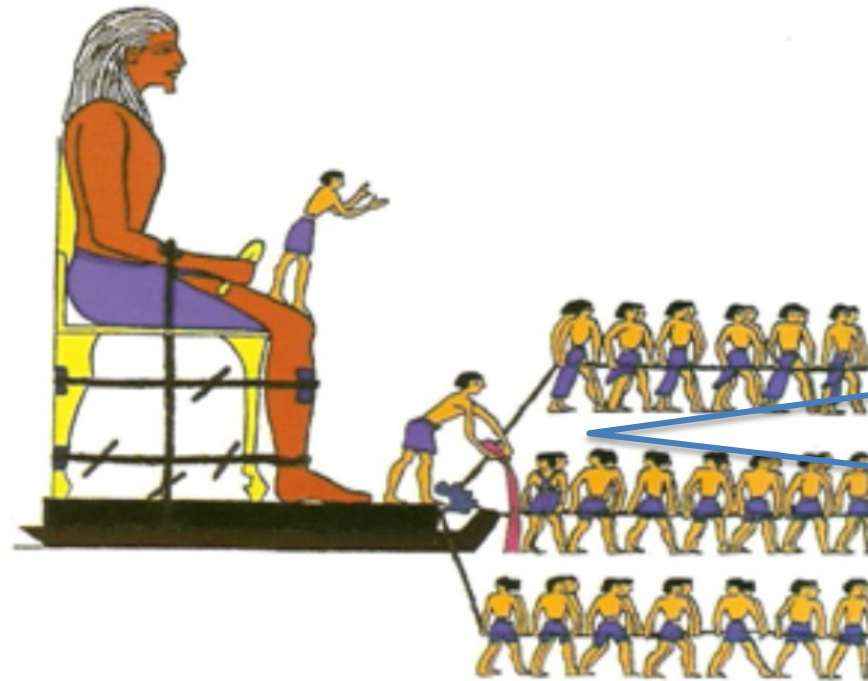
# Tribology

- from the Greek word τριβω meaning rubbing, literally tribology is the “**science of rubbing**”
- dictionaries define tribology as the **science and technology of interacting surfaces in relative motion**
- tribology includes the study of **friction, wear and lubrication**
- understanding these phenomena requires knowledge from **physics, chemistry, mechanics, materials science. Tribology is a truly interdisciplinary science.**
- the word tribology is recent, it was coined by the Jost committee in 1966, but the interest in tribology is much older



# interest in tribology older than recorded history

records show the use **wheel from 3500 BC**: our ancestors were concerned with reducing friction in sliding motion



**the first recorded tribologist (2000 BC)** is poring a liquid, most likely water, into the path

during and after the Roman empire tribology principles were mostly used to design war **machinery and fortification**



# Leonardo da Vinci: a precursor

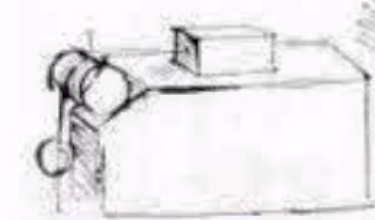


**Da Vinci**  
**1452-1519**

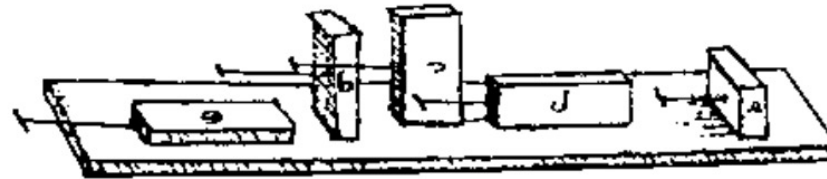
**Leonardo da Vinci** was the first to study friction systematically

two phenomenological laws of friction:

1. if the load of an object is doubled, its friction will also double



2. the areas in contact have no effect on friction



- **Da Vinci laws 200 years before Newton even defined what a force is**
- **his work had no influence on subsequent studies, because his notes remained unpublished for hundred of years**



# classical friction laws

in **1699 Amontons** rediscovered the two basic rules of friction:

1.  $F = \mu F_N$  Amontons' law
2.  $F$  independent from the contact area



**Amontons**  
**1663-1705**

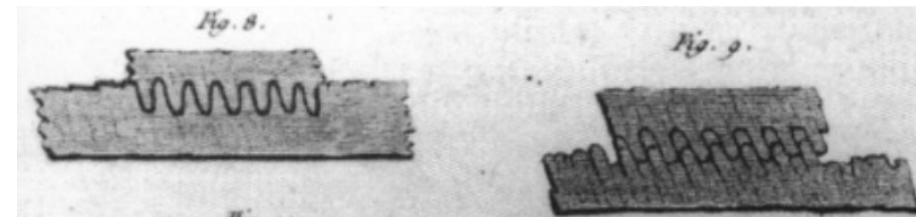


**Coulomb**  
**1736-1806**

**Coulomb (1785)** in the *Theorie des machines simples* made a distinction between **static and kinetic friction**

3.  $F = a + b \cdot \log(v)$  in the first hours of motion, then the friction force is independent from the sliding velocity

4. friction due to interlocking of asperities



# real area of contact

**For several centuries scientists believed that friction was due to the roughness. Geometrical hypothesis of friction.**

**Desanguliers (1734) first proposed that **adhesion** as a key element in the friction process. Tribologists rejected the idea as it appeared to contradict the independence on the contact area**

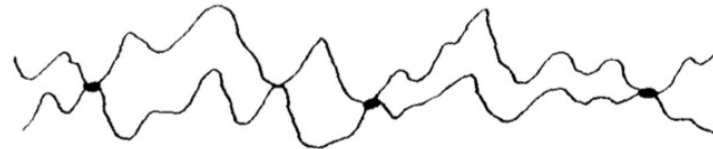


**Bowden  
1903-1968**



**Tabor 1913-  
2005**

**Contradiction cleared up by the introduction of the concept of **the real area of contact (1950)****

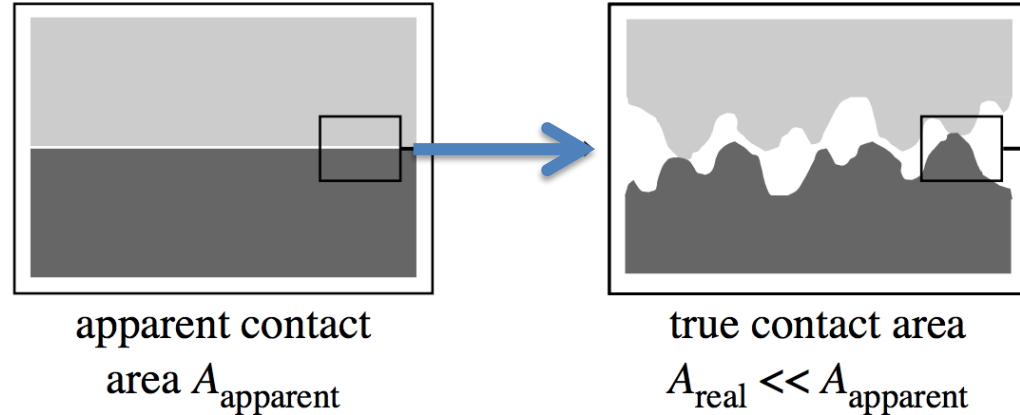


"putting two solids together is rather like turning Switzerland upside down and standing it on Austria – their area of intimate contact will be small"

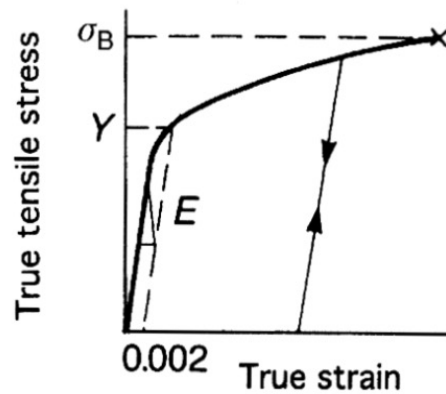


# asperity contact

real area of contact  
made up of a large  
number of small  
regions of contact,  
asperities or junctions



## Strength of materials



$$A_r = \sum A_i$$

$A_r$  increases with  $F_N$  until the real area of contact is just sufficient to support the load. For plastic contacts:

$$A_r = F_N / \tau_Y$$

$\tau_Y$  Yield strength: the maximum resistance to plastic deformation



# adhesion theory of friction

Bowden and Tabor proposed that the friction force is necessary to overcome adhesion at asperity contacts

F due to adhesion  $F = \tau_s A_r$

$\tau_s$  shear strength: maximum resistance to sliding

$F = \tau_s A_r = \tau_s / \tau_Y F_{N,r}$   $\mu = \tau_s / \tau_Y$

consistent with the two Amontons laws...

...however things are more complicated:

- $\tau_Y$  reduced by the shear force
- $\tau_s$  can depend on the contact pressure  $\tau_s = \tau_0 + a p$
- for elastic contacts  $F \sim (F_N)^{2/3}$
- **Archard (1953)** load-dependent number of asperities in contact -> no contradiction with the Amontons 1<sup>st</sup> law



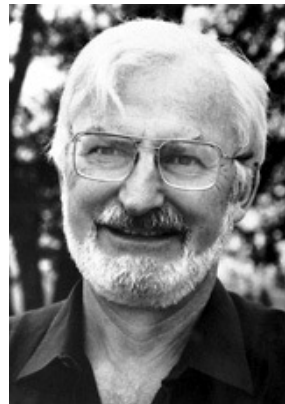
# nanotribology: the renaissance of friction

No progresses in the understanding of friction until '80, when an exciting era of renewed interest in tribology has started. The fueling factors are:

1. advent of **new experimental techniques** that allow to probe interfacial properties with atomic resolution

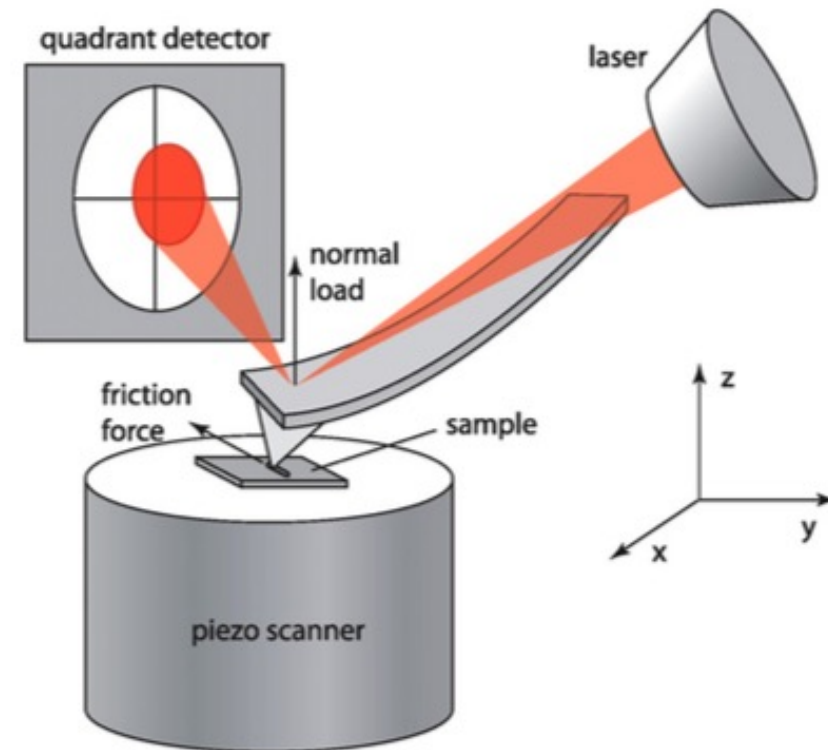


*Gerd Binnig*

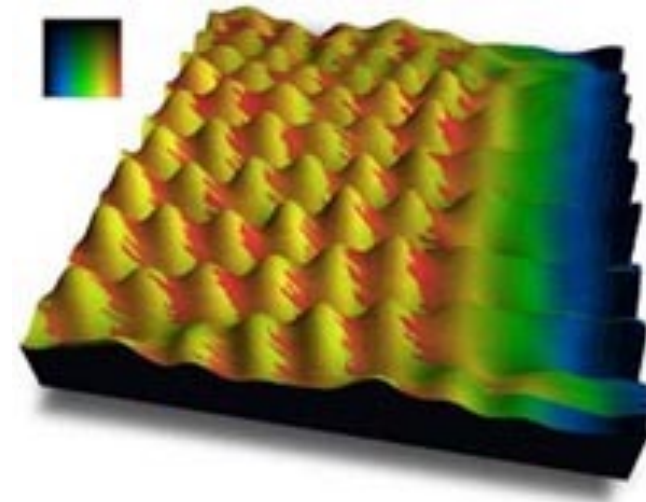
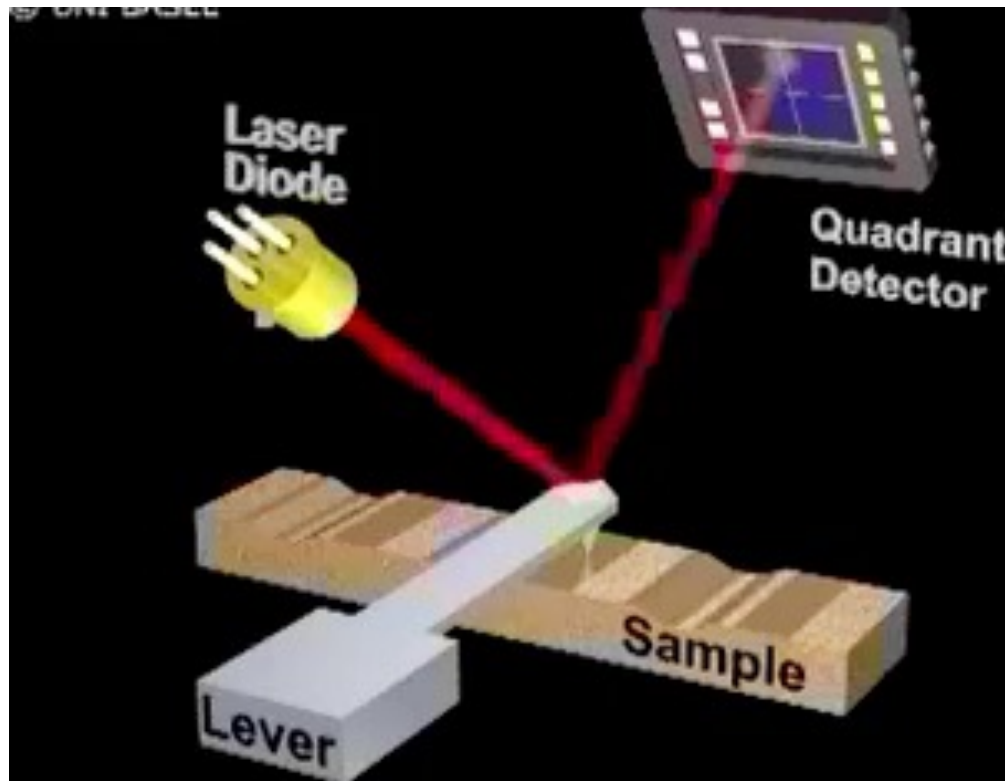


*Heinrich Rohrer*

**1986, Nobel Prize in Physics for scanning tunneling microscopy (STM)**

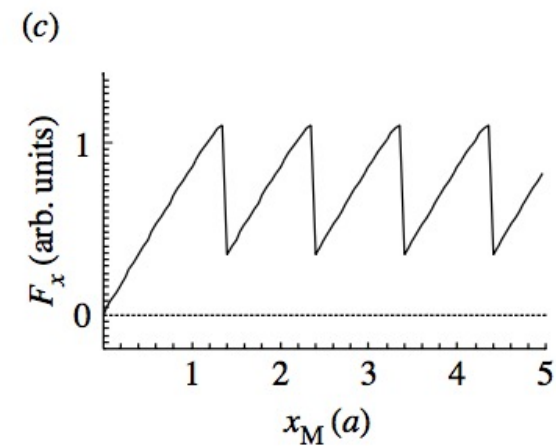
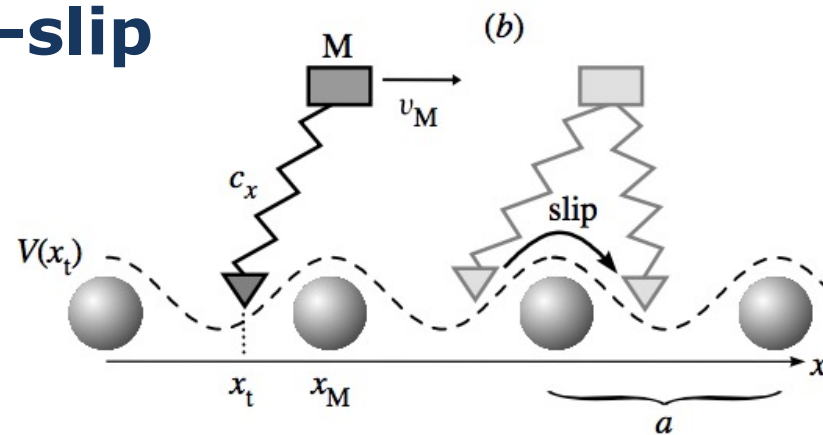






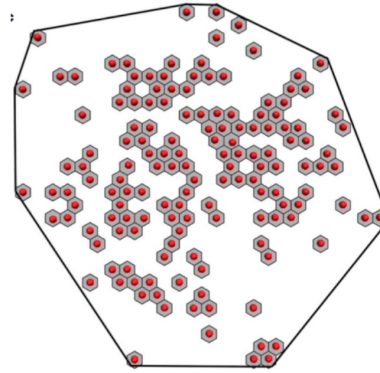
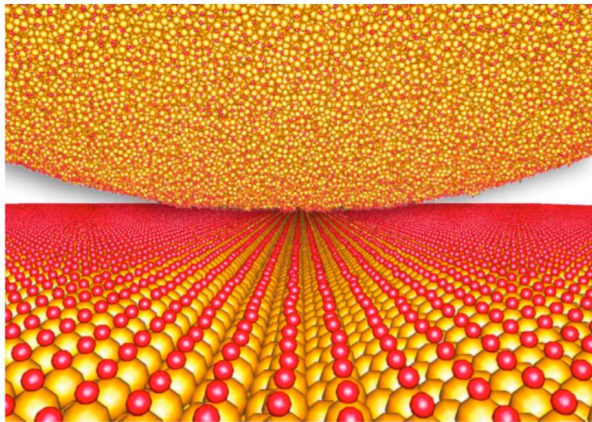
**2.5 x 2.5 nm simultaneous topographic and friction image of HOPG.**

## stick-slip



## 2. advances in computational methods and computer power.

Atomistic simulations, molecular dynamics, to model the asperity during sliding:

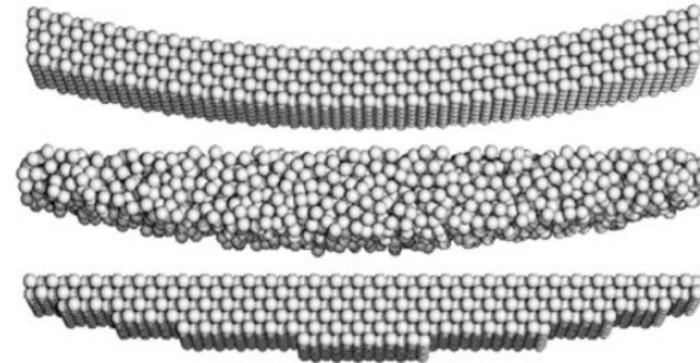


$$A_{\text{real}} = N_{\text{at}} A_{\text{at}}$$

Y. Mo, K. T. Turner  
and I. Szlufarska,  
*Nature* 457, 1116  
(2009)

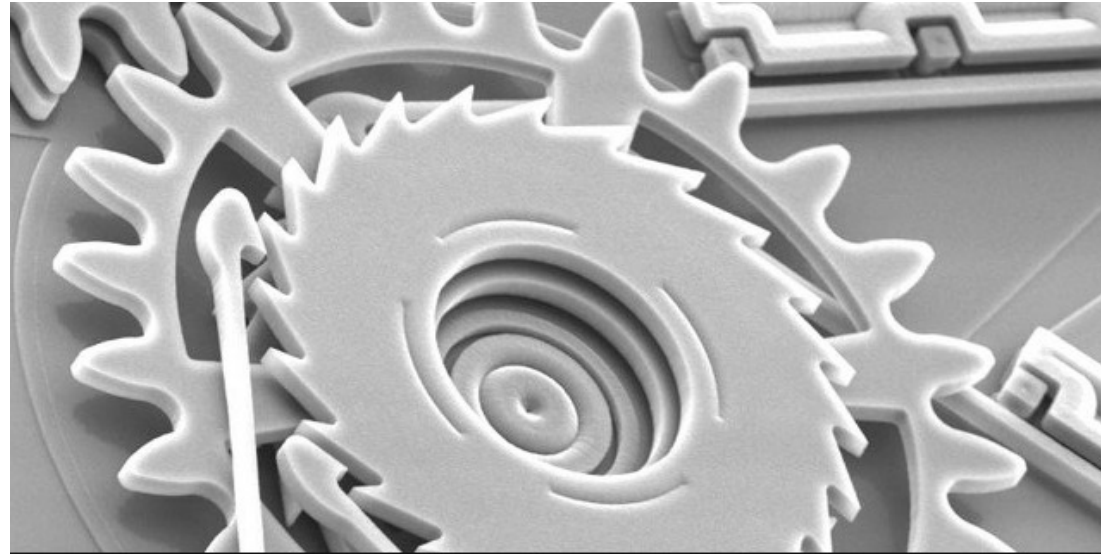
The surface roughness produced by discrete atoms leads to dramatic deviations from continuum theory.

B. Laun and M. O. Robbins, *Nature* 435, 929 (2005)



### 3. Nanotechnology

miniaturized devices with moving components, with **high surface/volume ratio** suffer problems of adhesion and friction



Micro-, nanoelectromechanical systems (MEMS & NEMS)

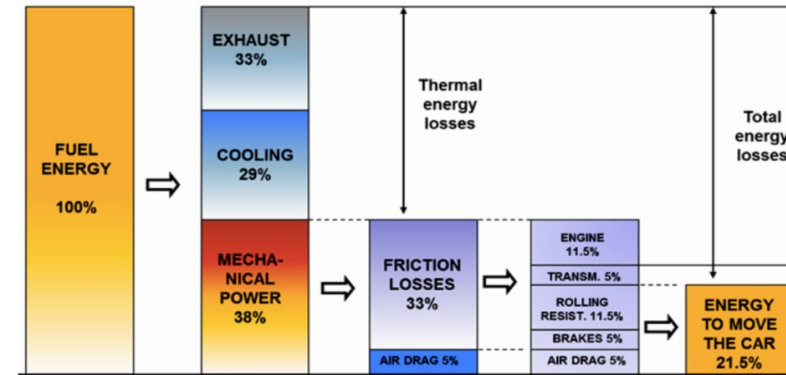
### 4. Energy saving

approximately 1/3 of the world's energy is lost in frictional processes



# Impact of tribology on energy and environment

- Friction and wear cause massive energy and environmental costs



K. Holmberg and A. Erdemir, FME Transactions 43, 181 (2015)

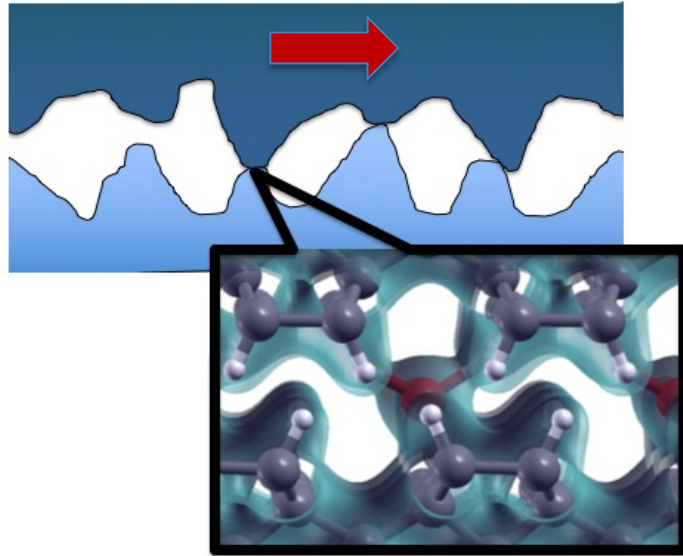
10% less friction in cars = fuel saving ( $10^{11}$  lt/yr) & less CO<sub>2</sub> emissions ( $10^{11}$  Kg/yr)



Advancing the technologies to reduce friction:  
**Materials**



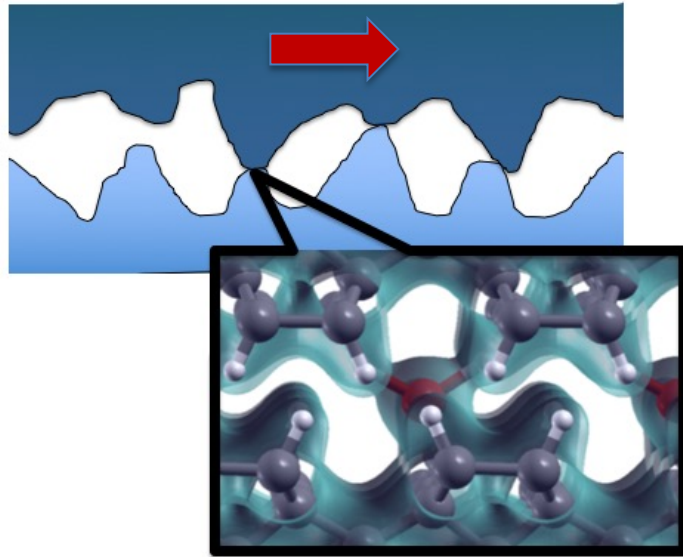
# Ab initio MD to open a window on the buried sliding interface



- **Simulations** can open a window on the sliding buried interface
- **Quantum mechanics essential** to accurately describe asperity adhesion and tribochemical reactions at **nanosized contacts**

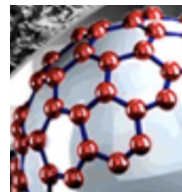


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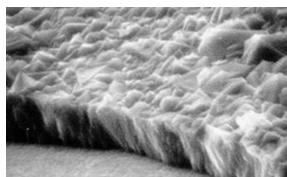


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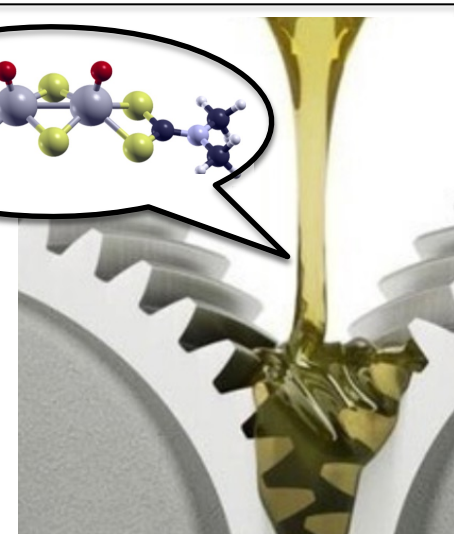
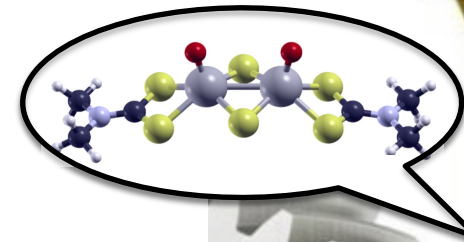
✓ **Solid lubrication:**  
surface covered  
by inert layers or  
thin films



DLC



✓ **Boundary lubrication:**  
“tribofilms”  
formed *in situ* by  
tribochemical  
reactions

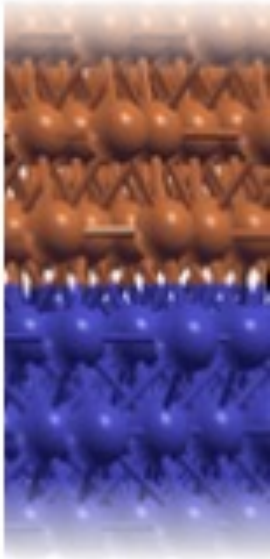
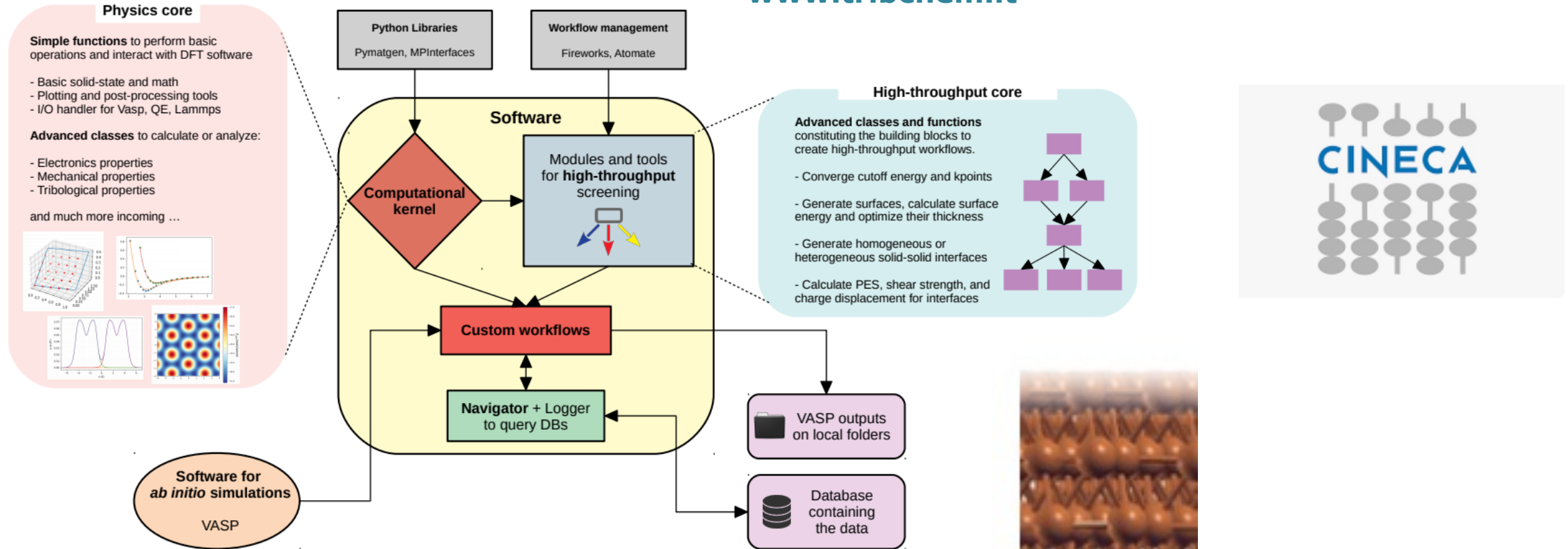


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# TRIBCHEM for high throughput calculations of solid interfaces

[www.tribchem.it](http://www.tribchem.it)



G. Losi, O. Chehaimi, and M.C. Righi,  
Journal of Chemical Theory and Computation (2023)

New release coming soon!





# First principles calculation of adhesion and shear strength

## work of adhesion

energy per unit area required to separate two surfaces from contact

$$E_{adh} = (E_{12} - E_1 - E_2)/A$$

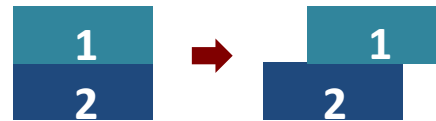
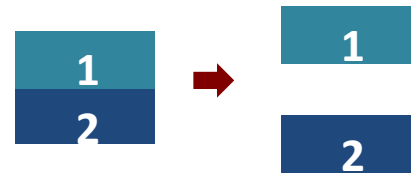
## stacking fault energy surface

potential energy surface (**PES**) for the sliding interface  $E_{adh}(x,y)$

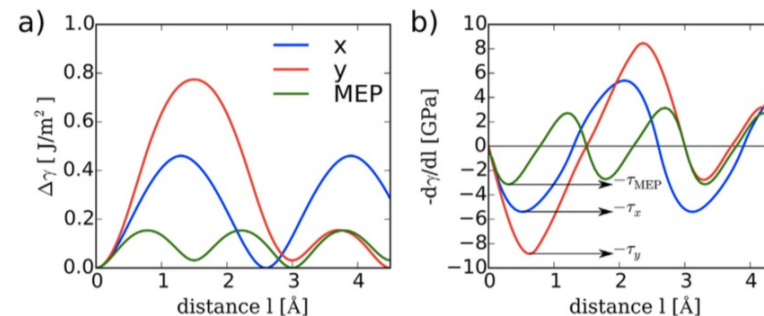
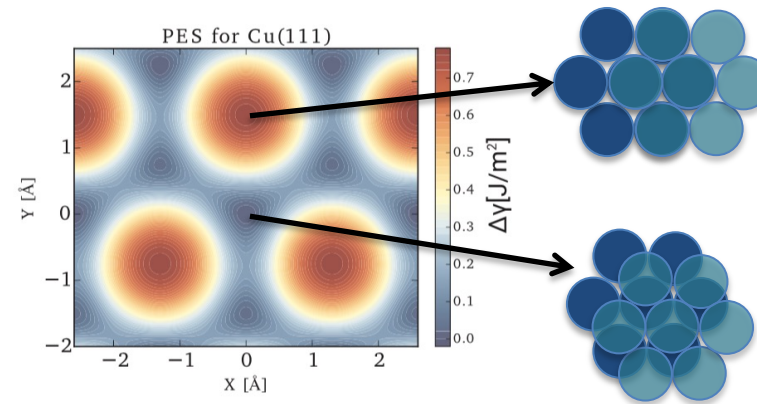
## ideal shear strength

maximum restoring force

$$\tau = \max | - \nabla E_{adh}(x,y) |$$



G. Zilibotti and M. C. Righi,  
Langmuir 27, 6862 (2011)



# Database for the adhesion of metallic interfaces

**Adhesion Energy (J/m<sup>2</sup>)**

● W(110)	-5.87	-4.54	-5.50	-4.29	-5.38	-4.51	-4.49	-4.70	-3.44	-3.96	-2.15	-2.66	-2.39	-2.38
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● Ir(111)	-5.38	-5.27	-5.26	-4.21	-4.41	-3.65	-5.16	-3.41	-2.99	-3.66	-2.02	-2.08	-2.71	-2.37
● Ni(111)	-4.51	-3.27	-3.95	-4.12	-3.65	-3.74	-4.39	-3.29	-2.75	-3.32	-2.05	-2.17	-2.39	-2.17
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● Au(111)	-2.66	-2.54	-2.51	-2.61	-2.08	-2.17	-3.47	-1.94	-1.80	-1.79	-1.56	-1.33	-2.00	-1.24
● Mg(001)	-2.39	-2.23	-2.36	-2.10	-2.71	-2.39	-1.99	-3.18	-1.86	-1.19	-1.41	-2.00	-1.21	-0.97
● Zn(001)	-2.38	-2.32	-2.33	-2.49	-2.37	-2.17	-2.22	-2.27	-1.53	-0.81	-1.08	-1.24	-0.97	-0.89
	W(110)	Cr(110)	Mo(110)	Fe(110)	Ir(111)	Ni(111)	Ti(001)	Pt(111)	Cu(111)	Al(111)	Ag(111)	Au(111)	Mg(001)	Zn(001)

P. Restuccia, O. Chehaimi, G. Losi, M. Marsili and M.C. Righi  
 ACS Advanced Materials Interfaces 15, 19624 (2023)



# adhesion of metallic interfaces predicted by ML

The sure independent screening and sparsifying operator (SISSO):

- extracts effective materials descriptors out of a number of possibly correlated features
- identifies an analytical equation able to describe the descriptor relationship

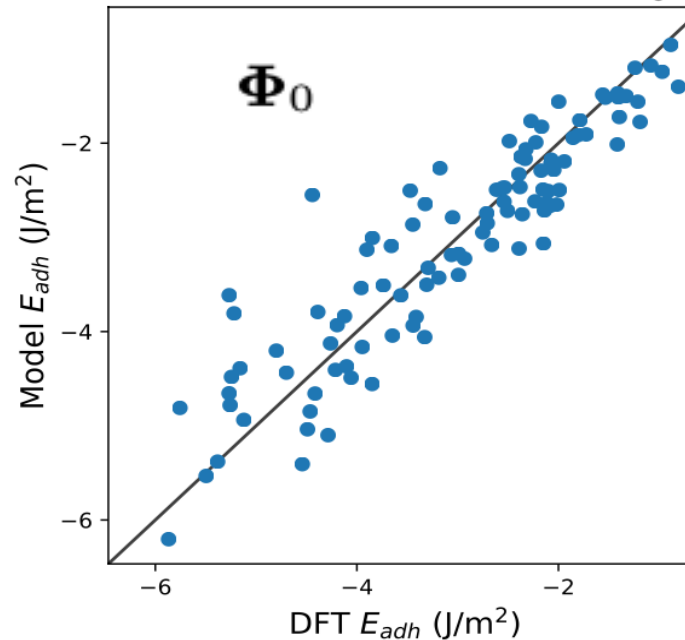
Ouyang, S. Curtarolo, E. Ahmetcik, M. Scheffler, and L. M. Ghiringhelli, *Phys. Rev. Materials* 2, 083802 (2018)

Screened descriptors:

$\epsilon_{1,2}$  cohesive energy,  $K_{1,2}$  bulk modulus,  $\gamma_{1,2}$  surface energy,  $\rho_{1,2}$  atom density at surf.,  $(e_1 - e_2)$  electronegativity difference their arithmetic (AM) and geometric (GM) averages

$$E_{adh} = A\sqrt{\gamma_1\gamma_2} + B\sqrt{\epsilon_1\epsilon_2} + C\sqrt{K_1K_2}$$

$$A = -1.13, B = -0.34; C = 0.0016$$

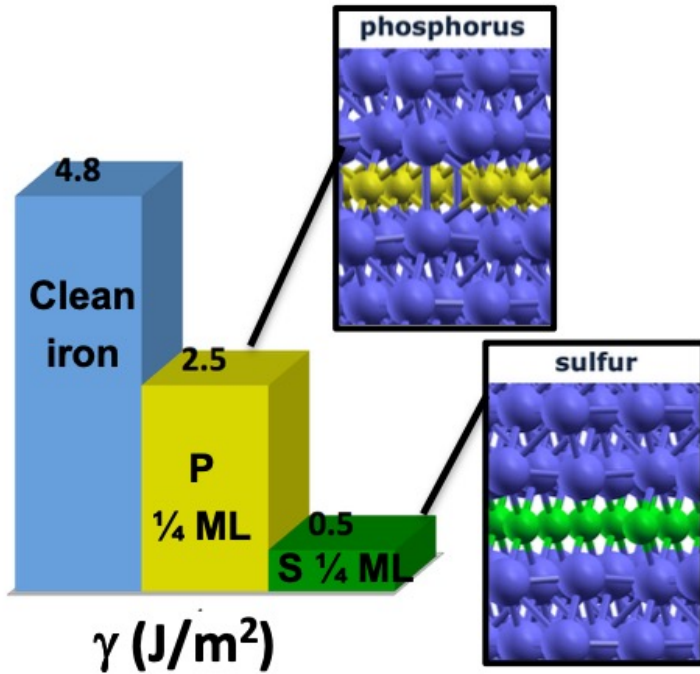


Model	$R^2$	RMSE (J/m <sup>2</sup> )
$\Phi_0$	0.83	0.52

P. Restuccia, O. Chehaimi, G. Losi, M. Marsili and M.C. Righi, *ACS Advanced Materials Interfaces* 15, 19624 (2023)



# Effects of surface chemical modifications on interfacial adhesion

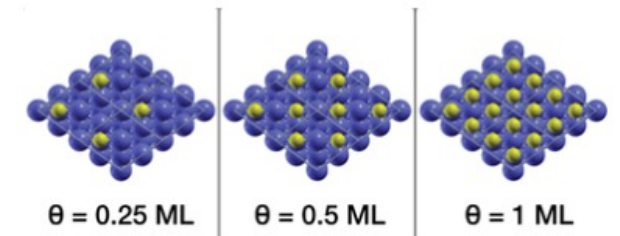


adhesion  
reduced by  
chemical  
surface  
modifications



screening intercalated  
species for selected levels  
of adhesion and friction

G. Fatti, P. Restuccia, C. Calandra and M. C. Righi  
*J. Phys. Chem. C* 122, 28105 (2018)





# Effects of surface chemical modifications on metal/metal coverage

coverage

25%

50%

100%

Elem	B	C	F	H	N	O	P	S	B	C	F	H	N	O	P	S	B	C	F	H	N	O	P	S	Clean
W	-6.36	-6.48	-0.92	-6.20	-5.42	-4.42	-4.23	-3.20	-6.60	-7.09	-1.19	-4.86	-5.41	-2.63	-2.40	-1.94	-8.03	-9.58	-2.99	-3.43	-5.05	-2.17	-3.65	-1.86	-5.87
Cr	-5.43	-5.20	-2.59	-5.94	-5.88	-4.89	-3.13	-2.93	-6.06	-7.61	-0.95	-5.79	-6.39	-3.90	-2.85	-2.29	-8.59	-10.49	-2.07	-3.78	-1.23	-3.29	-4.11	-2.16	-5.76
Mo	-5.54	-5.84	-2.05	-5.20	-5.22	-4.09	-3.56	-3.13	-6.15	-6.67	-1.31	-4.40	-5.39	-2.51	-2.85	-2.14	-7.76	-9.34	-3.32	-3.40	-4.73	-2.45	-3.68	-1.90	-5.12
Fe	-5.05	-5.35	-2.20	-4.82	-4.71	-3.54	-3.22	-2.36	-2.32	-6.17	-0.66	-4.79	-4.81	-2.71	-2.91	-1.51	-8.00	-8.34	-1.98	-3.23	-3.80	-2.66	-3.18	-1.48	-4.80
Ir	-4.09	-3.64	-0.24	-3.53	-2.47	-1.89	-2.32	-0.71	-4.43	-3.60	-0.45	-2.81	-2.62	-0.04	-2.20	-0.95	-7.32	-5.06	-1.39	-1.70	-0.56	-0.67	-2.75	-1.24	-4.41
Ni	-4.29	-4.30	-0.98	-3.41	-3.46	-1.99	-4.38	-3.03	-5.99	-5.70	-2.67	-3.80	-3.99	-3.34	-3.88	-2.07	-7.43	-7.29	-2.65	-3.49	-5.22	-5.07	-3.87	-0.99	-3.47
Ti	-4.72	-5.54	-3.11	-4.37	-4.89	-4.79	-4.19	-3.49	-5.78	-6.93	-2.09	-3.55	-6.26	-5.16	-4.43	-3.20	-8.19	-12.02	-0.00	-2.80	-10.08	-7.05	-5.17	-4.03	-3.33
Pt	-3.19	-2.68	-0.32	-2.37	-1.55	-1.06	-1.92	-0.95	-4.01	-2.99	-0.52	-1.81	-2.20	-1.04	-2.29	-1.12	-6.99	-3.84	-1.76	-0.85	-0.42	-0.65	-2.12	-1.15	-2.99
Cu	-2.86	-2.11	-0.48	-2.08	-1.65	-1.40	-1.08	-0.56	-4.08	-2.53	-0.92	-1.61	-1.91	-1.43	-1.19	-0.81	-3.35	-4.08	-3.02	-0.93	-0.42	-3.24	-1.04	-0.85	-2.54
Al	-2.41	-2.17	-0.05	-1.37	-1.41	-0.88	-0.79	-0.18	-3.18	-3.54	-0.03	-1.13	-2.36	-1.17	-0.98	-0.30	-4.22	-7.31	-0.01	-0.68	-7.34	-0.04	-1.05	-0.42	-1.42
Ag	-1.63	-1.37	-0.81	-1.34	-1.31	-1.09	-0.75	-0.48	-2.45	-2.39	-0.87	-1.08	-1.54	-1.32	-1.31	-0.77	-2.75	-2.73	-2.67	-0.87	-1.75	-2.35	-1.08	-0.92	-1.41
Au	-1.65	-1.10	-0.29	-0.84	-0.58	-0.75	-0.63	-0.10	-2.58	-1.50	-0.74	-0.49	-0.99	-0.94	-0.84	-2.09	-3.66	-0.64	-2.19	-0.06	-1.83	-2.47	-0.58	-0.43	-1.33
Mg	-2.02		-0.82	-1.12		-1.63	-1.33	-0.95	-2.80		-0.45	-1.05		-2.55	-1.58	-1.36	-4.09	-5.96	-0.01	-0.85			-2.05	-2.73	-1.21
Zn	-1.65	-1.58	-0.18	-0.32	-1.37	-0.87	-0.83	-0.58	-2.31	-2.40	-0.57	-0.84	-2.23	-1.98	-0.83	-1.16	-2.57	-2.63	-0.58	-0.07	-3.15	-4.15	-0.58	-0.37	-0.89

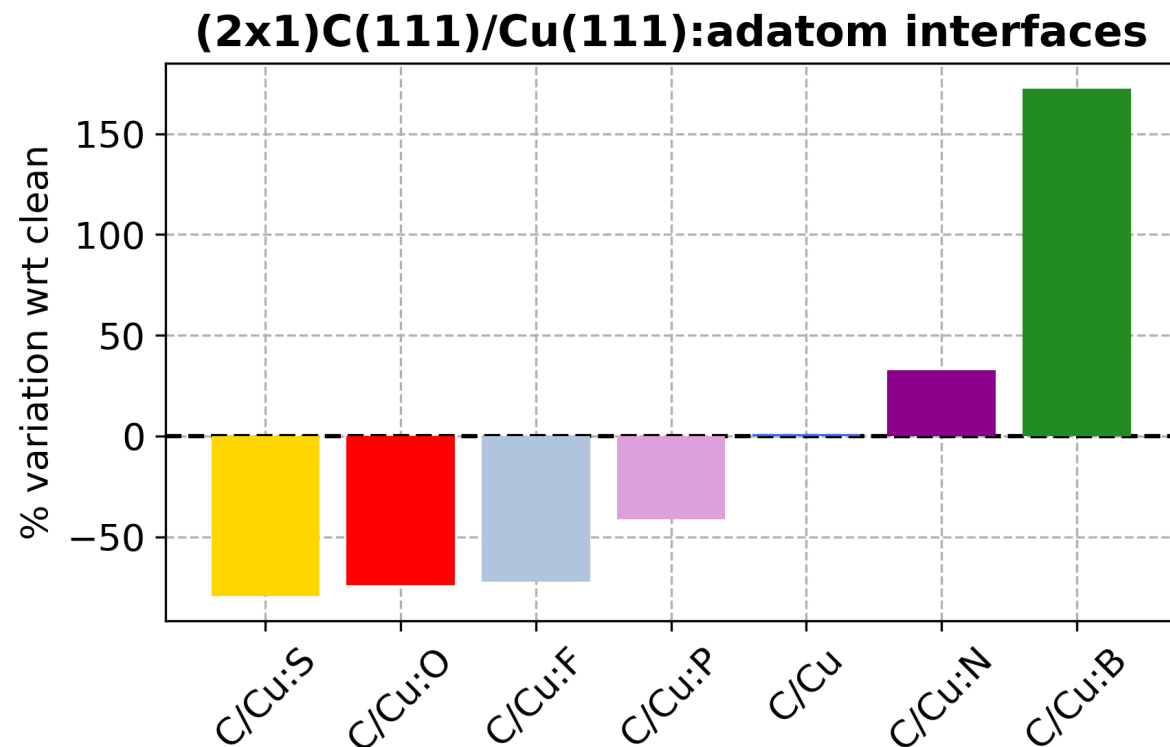
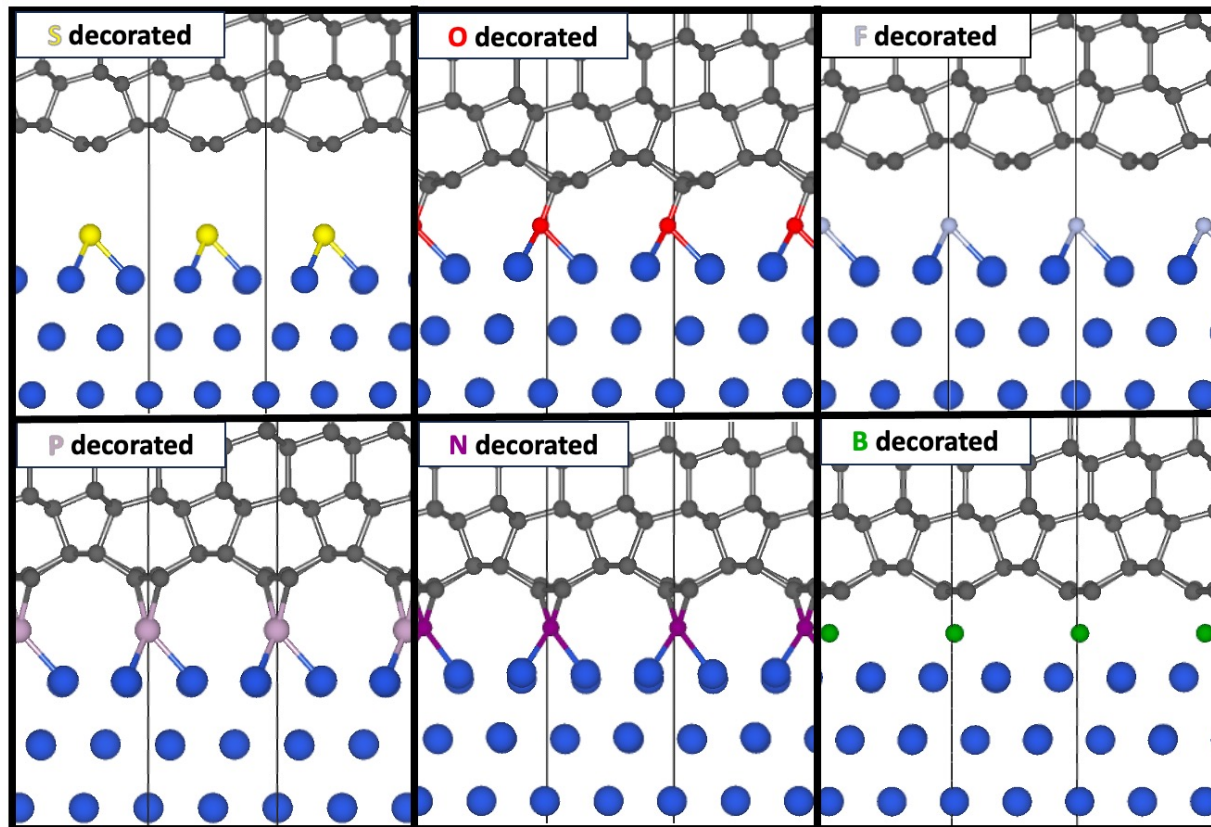
Adhesion (J/m<sup>2</sup>)

- P, S and in particular F are the strongest adhesion reducers
- B and C and sometimes N are the adhesion enhancers

E. Poli, M. Cutini, M. A. Nosir, O. Chehaimi and M. C. Righi, Applied Surface Science (2024)



# Effects of surface chemical modifications on covalent/metall



- **B and N act as adhesion enhancers** → adhesion increasing of 172% and 33%, respectively
- **S makes the Cu surface completely inert** → adhesion reduction of 80%

E. Damiani, M. Marsili and M. C. Righi, to be published (2024)



- **Designing Materials by High Throughput Calculations**
- **Modeling Materials Function by Molecular Dynamics**
- **What is tribology ?**
- **High throughput calculations to design solid interfaces**
- **Molecular dynamics to unravel mechanism of function of lubricants**
- **Simulating what happens in a reactor: H<sub>2</sub> and CNT production from CH<sub>4</sub>**

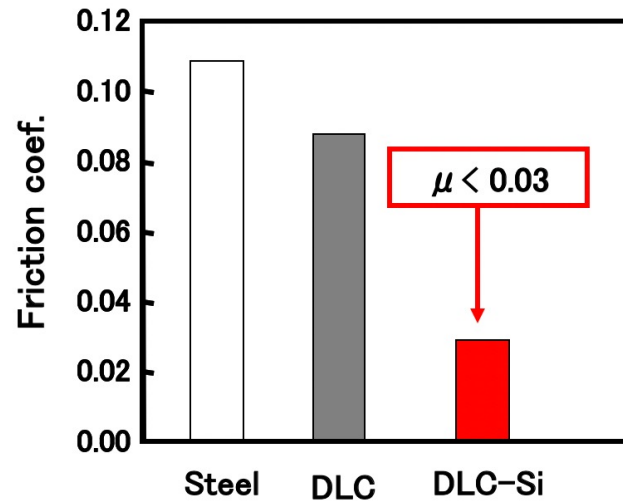


# Designing coatings

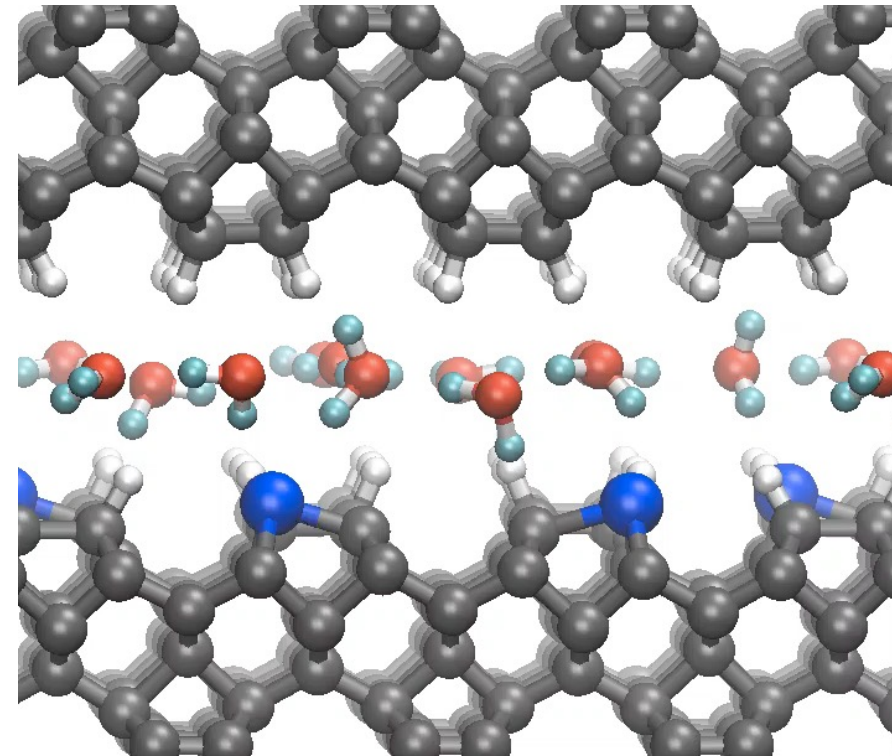


## diamond like carbon (DLC)

used in automotive and racing to coat valves and part of the piston



Silicon dopants increase the surface hydrophilicity and reduce friction





# 2D materials formed in situ by mechanochemistry

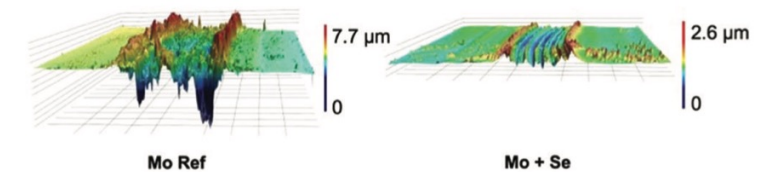
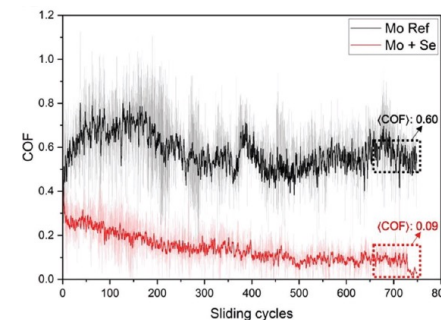
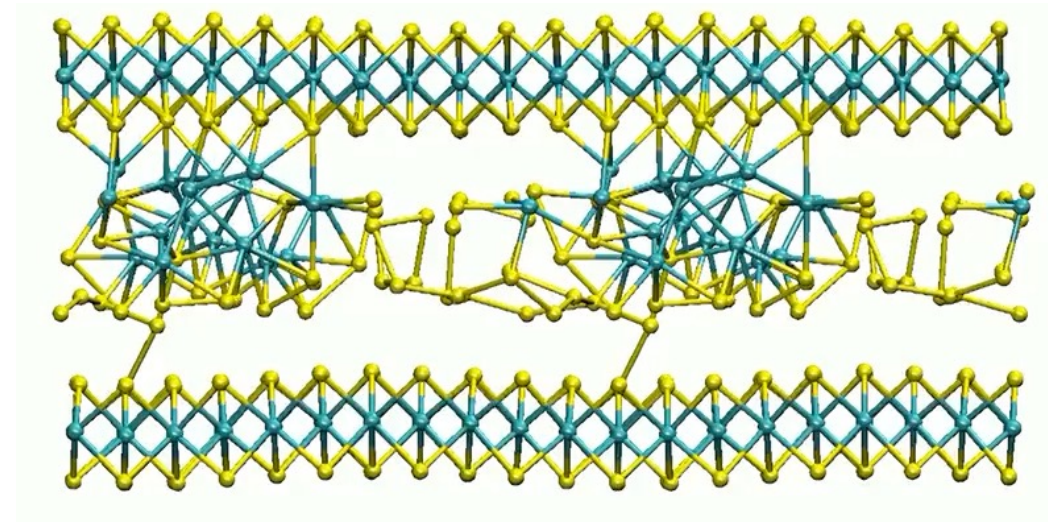
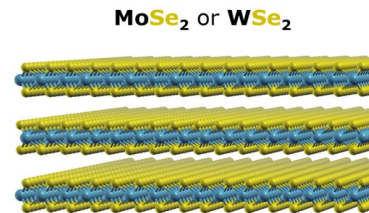
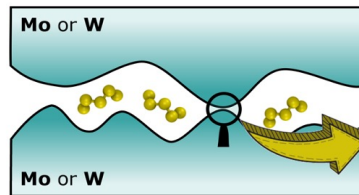
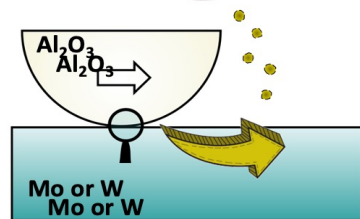
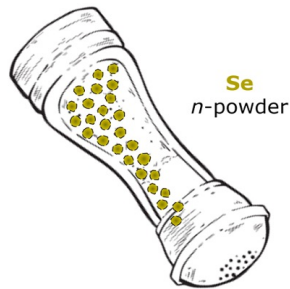
RESEARCH ARTICLE

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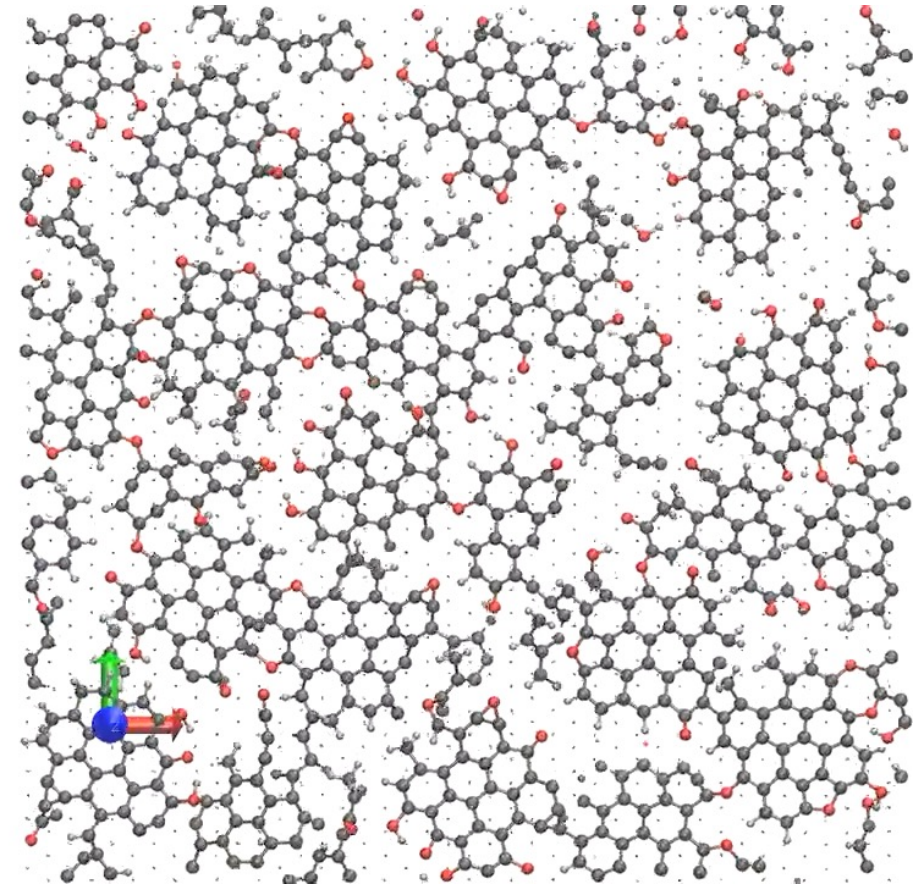
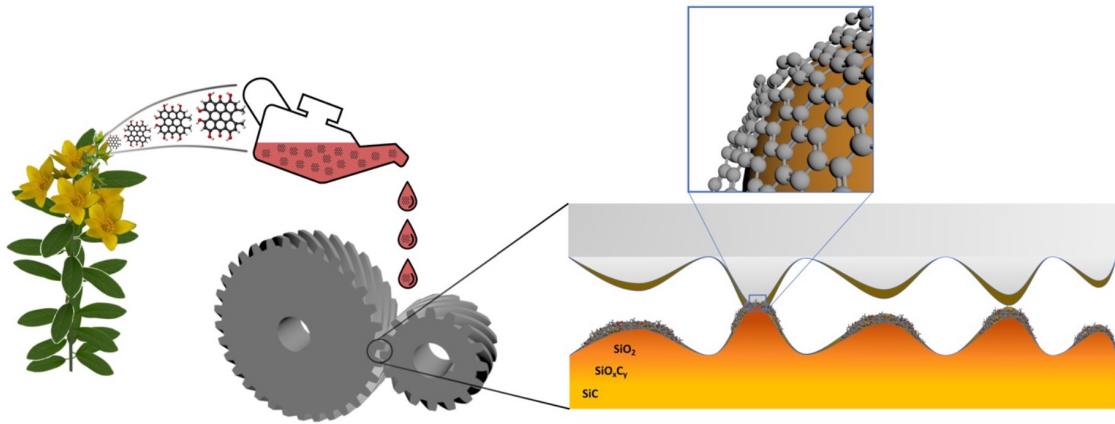
## Se Nanopowder Conversion into Lubricious 2D Selenide Layers by Tribochemical Reactions

Philipp G. Grützmacher,\* Michele Cutini, Edoardo Marquis, Manel Rodríguez Ripoll, Helmut Riedl, Philip Kutrowatz, Stefan Bug, Chia-Jui Hsu, Johannes Bernardi, Carsten Gachot,\* Ali Erdemir,\* and Maria Clelia Righi\*

Adv. Mater. 2023, 2302076



# 2D materials formed in situ by mechanochemistry



- Y. Long, A. Pacini, M. Ferrario, N. Van Tran, S. Peeters, B. Thiebaut, S. Loehl , J.M. Martin, M.C. Righi, and M.I. De Barros Bouchet, [Graphene-induced superlubricity through antiviral hypericin in glycerol](#). A new concept for green lubrication, **in printing Carbon (2024)**
- S. Peeters, G. Losi, S. Loehl  and M.C. Righi, [Aromatic molecules as sustainable lubricants explored by ab initio simulations](#), **Carbon 203, 717 (2023)**.

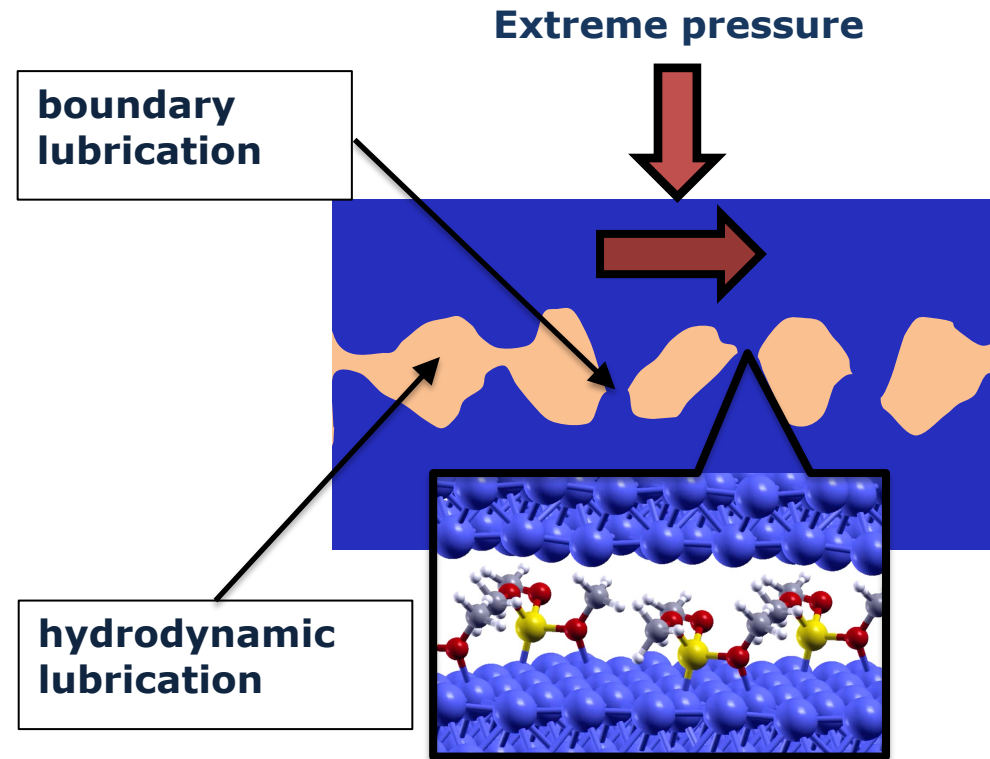
# Designing lubricant additives



Key elements for extreme-pressure additives are **sulfur** and **phosphorus**

Limits have been imposed due to their **harmful effect on the environment**

**Microscopic understanding** is essential to design new compounds



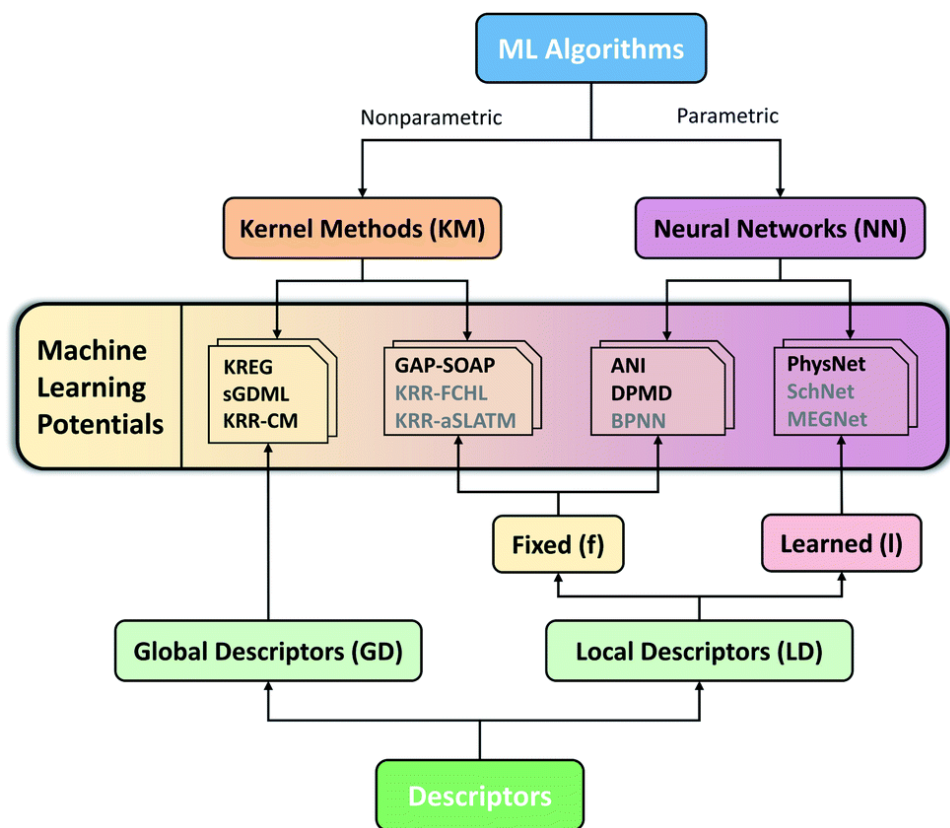


# Machine learning (ML) interatomic potentials

Accuracy vs efficiency dilemma:

describe atomic interactions with the accuracy of *ab initio* MD;

simulate system evolution with the efficiency of classical MD;



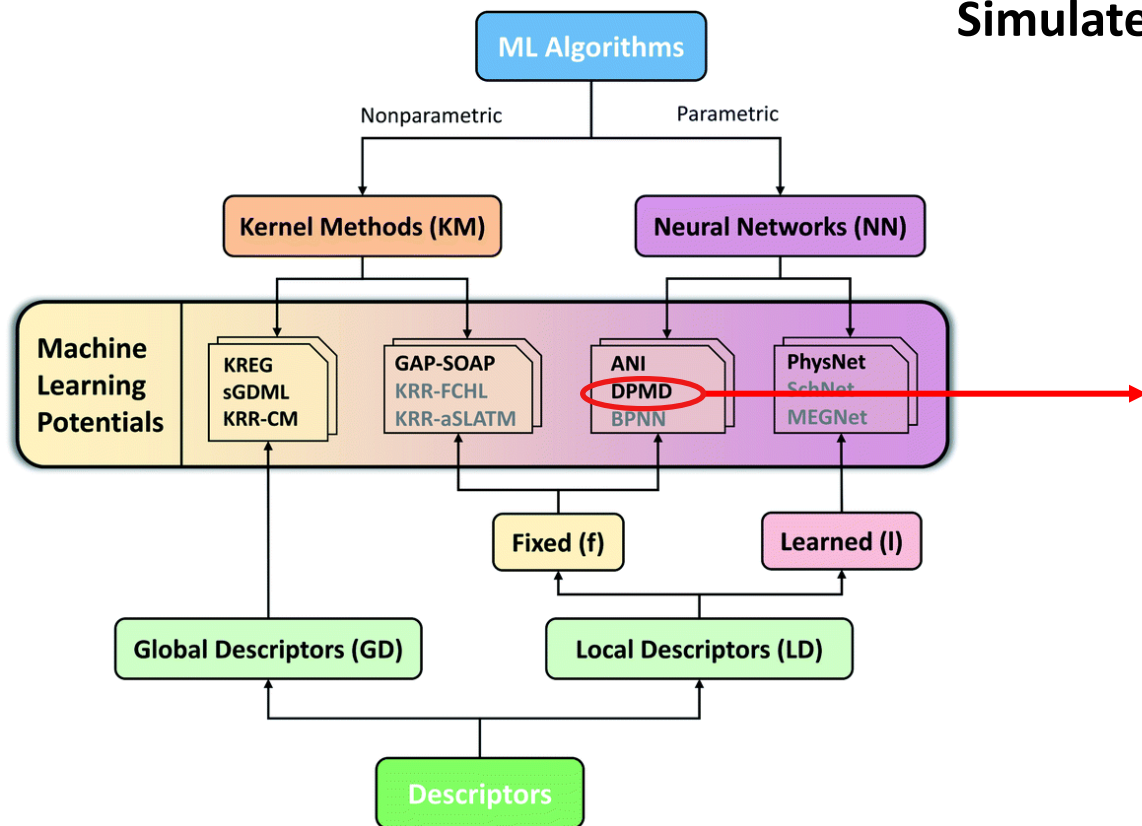
Ab initio data are used to fit or to train the **interatomic potentials** used in MD

# Machine learning (ML) interatomic potentials

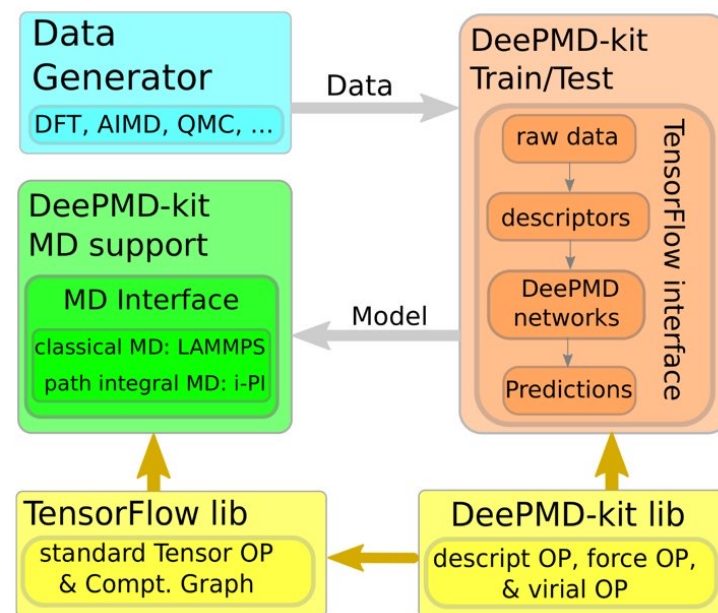
accuracy vs efficiency dilemma:

Describe atomic interactions with the accuracy of ab initio MD;

Simulate system dynamics with the efficiency of classical MD;



## DeePMD-kit



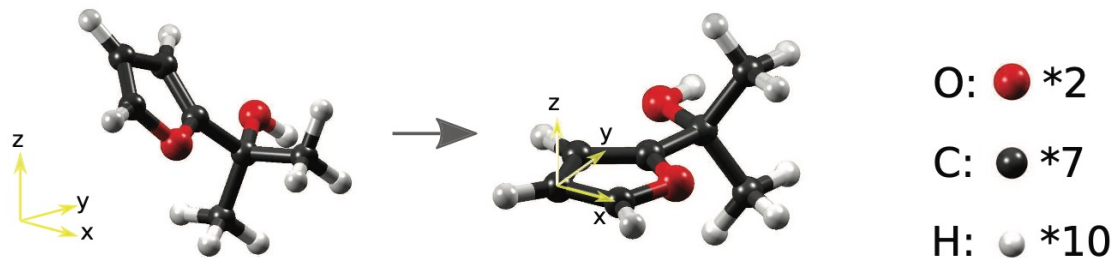
## Deep learning models based on convolutional NN

- L. Zhang, J. Han, H. Wang, R. Car, and W. E, *Phys. Rev. Lett.* **120**, 143001 (2018)
- Han Wang, Linfeng Zhang, Jiequn Han, and Weinan E. *Comp. Phys. Com.* **228**, 178 (2018).
- J. Zeng et al. *J. Chem. Phys.*, **159**, 054801 (2023).

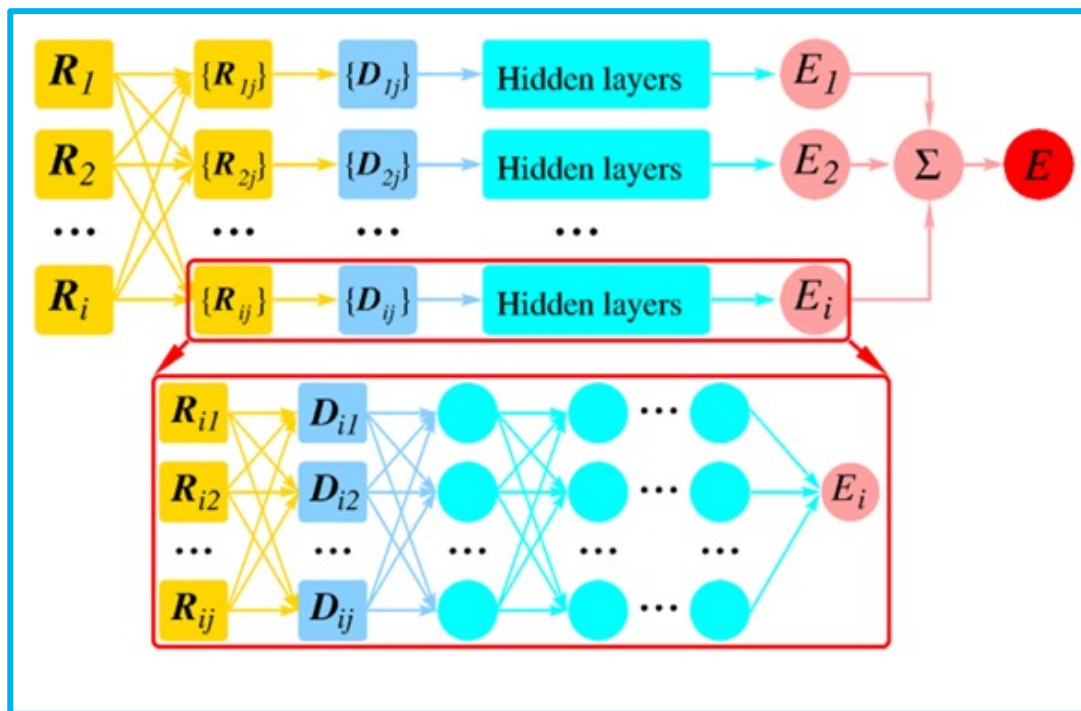
[M. Pinheiro, F. Ge, N. Ferré, P. O. Dral and M. Barbatti, *Chem. Sci.*, **12**, 14396 (2021)]



# DeePMD: method



## Neural Network



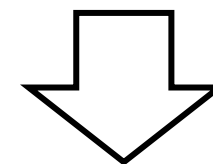
Descriptors define the local atomic environment of each atom within a cutoff distance

Descriptors are in-layer of a NN which has as out-layer the local atomic energy.

The total energy of the system configuration then obtained by summing up all the local energies.

$$L \propto \sum |E_{ab initio} - E_{NN}|^2 + |F_{ab initio} - F_{NN}|^2$$

Minimize the loss function adjusting the NN parameters



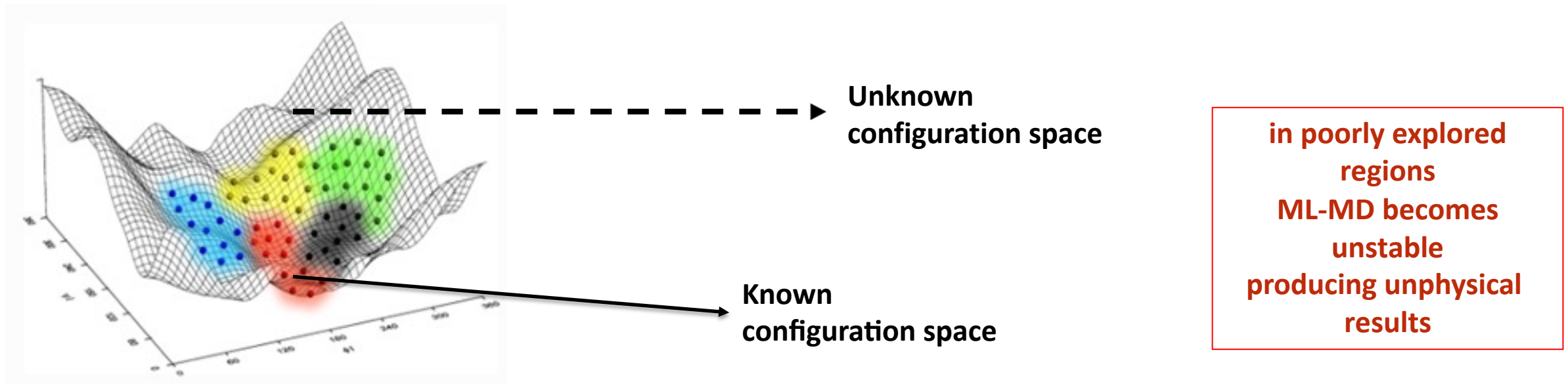
Train the Neural Network  
with ab initio data





# NN is an interpolator: unphysical results can be obtained if not well trained !

NN it produces a reliable dynamics when the systems evolves in regions of the configuration space sampled during the training

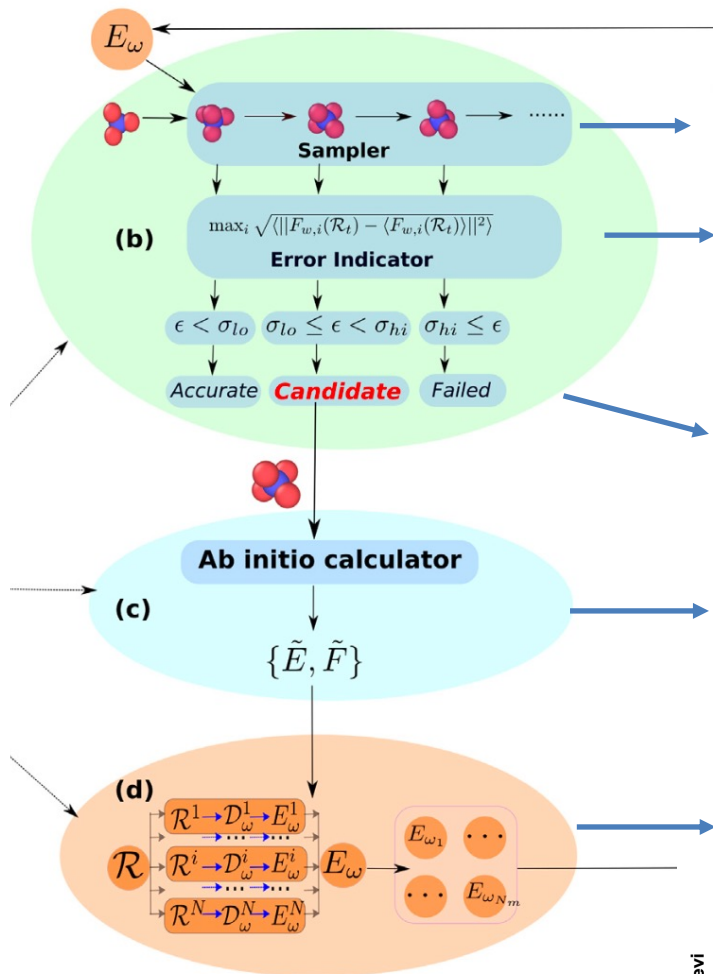


In the tribological systems the configuration space can vary a lot.  
In these systems a straightforward use of AIMD is not not sufficient



# Active learning

the atomic configurations used for the training are generated by the ML-MD itself, which reduces a lot the exploration time with respect to AIMD.



An initial dataset is used to train four NN

ML-MD with one NN to generate configurations, energy, forces

Calculate maximum force deviation produced by the four NN

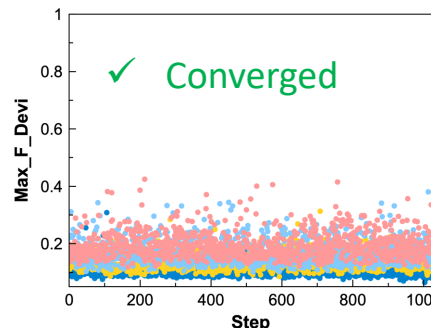
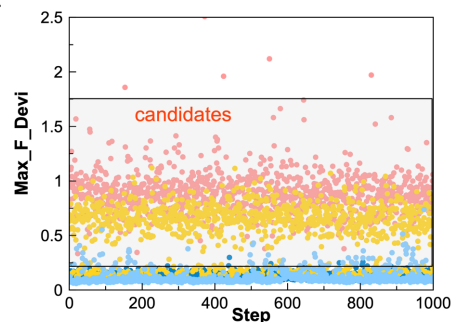
$$\epsilon_t = \max_i \sqrt{\langle \|F_{w,i}(\mathcal{R}_t) - \langle F_{w,i}(\mathcal{R}_t) \rangle\|^2 \rangle} \quad F_{w,i}(\mathcal{R}_t) = -\nabla_i E_w(\mathcal{R}_t)$$

$$\langle F_{w,i}(\mathcal{R}_t) \rangle = \frac{1}{N_m} \sum_{\alpha=1}^{N_m} F_{w,\alpha,i}(\mathcal{R}_t)$$

If the deviation is greater than a threshold the configuration is selected to enlarge the dataset

Ab initio calculations of energy and forces for selected candidates  
Dataset enlarged

Train 4 NNs with the new dataset.  
Repeat the procedure until no candidates are found



**Smart Configuration Sampling (SCS)** software developed by our group and coupled with QE

A. Pacini, M. Ferrario, S. Loehlè, M.C. Righi, in printing The European Physical Journal Plus (2024)

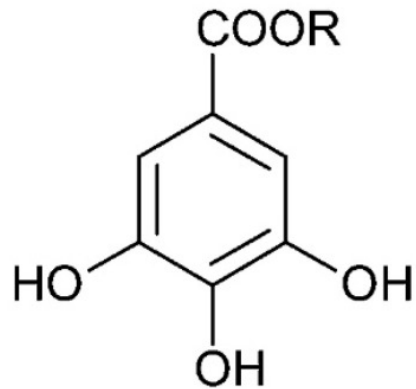
**Dp-gen package coupled with Vasp**

Yuzhi Zhang, Haidi Wang, Weijie Chen, Jinzhe Zeng, Linfeng Zhang, Han Wang, and Weinan E, Computer Physics Communications, 107206 (2020).



Train 4 models from the same data

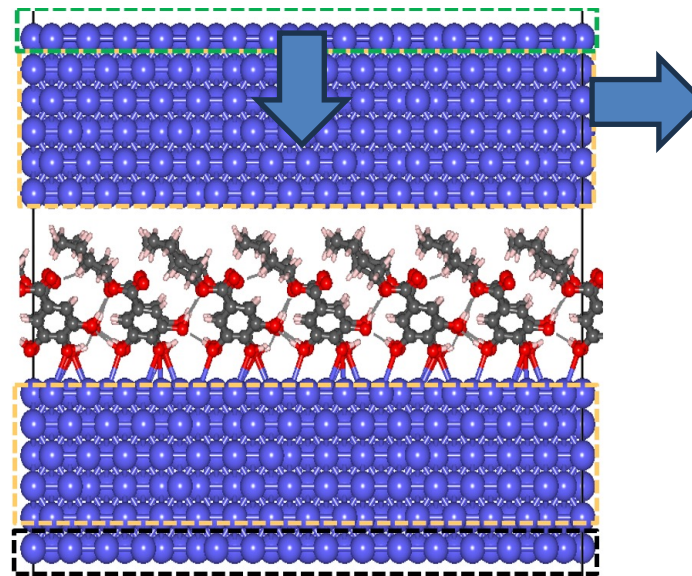
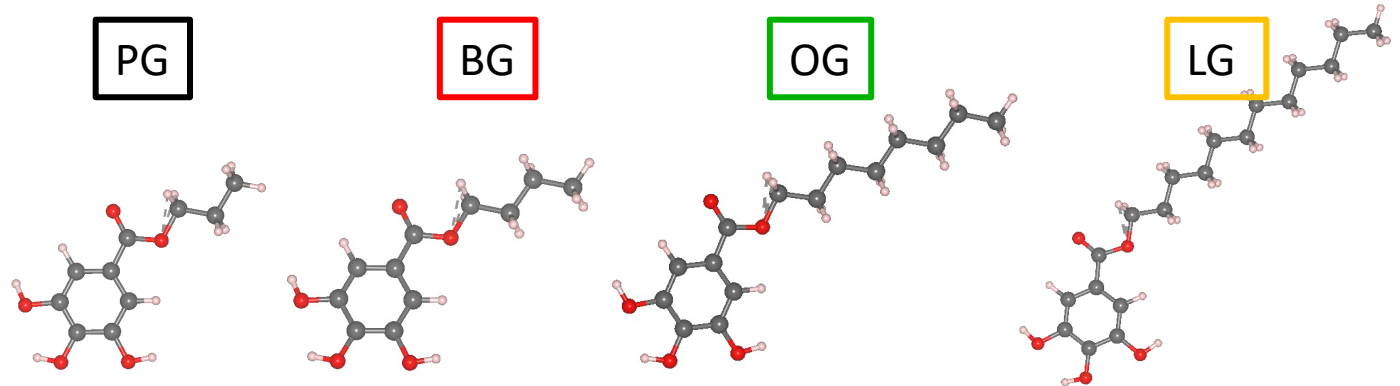
# NN potential for Gallate molecules at sling iron interfaces



Easily accessible

Anti-Oxidant

Eco-friendly



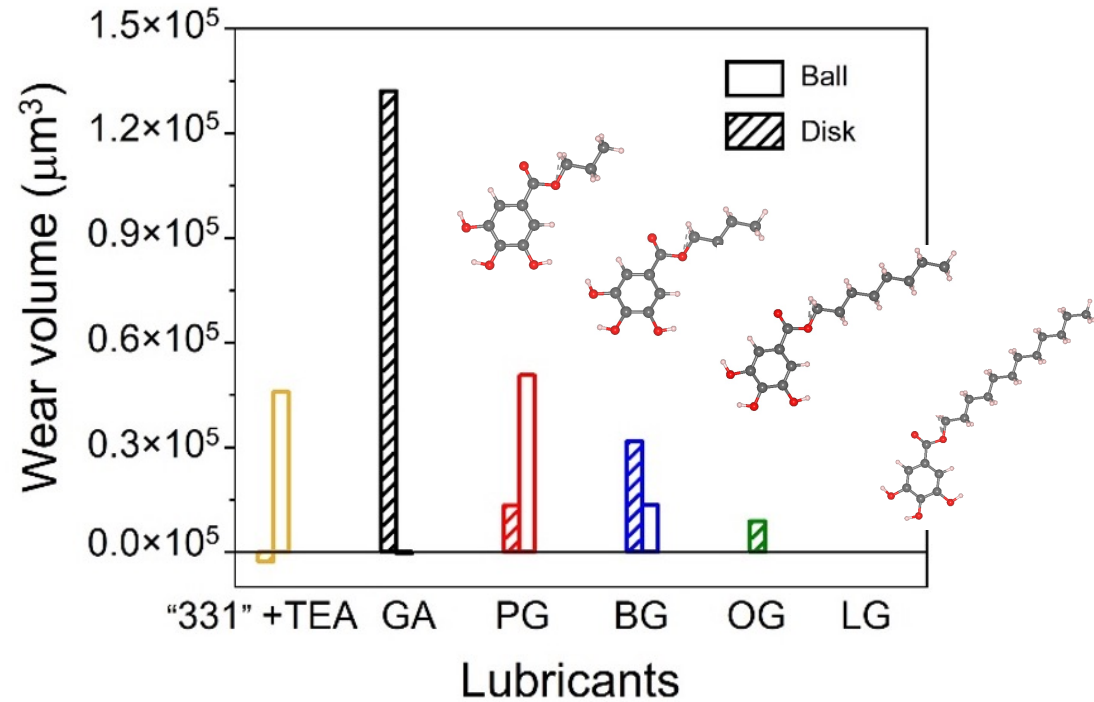
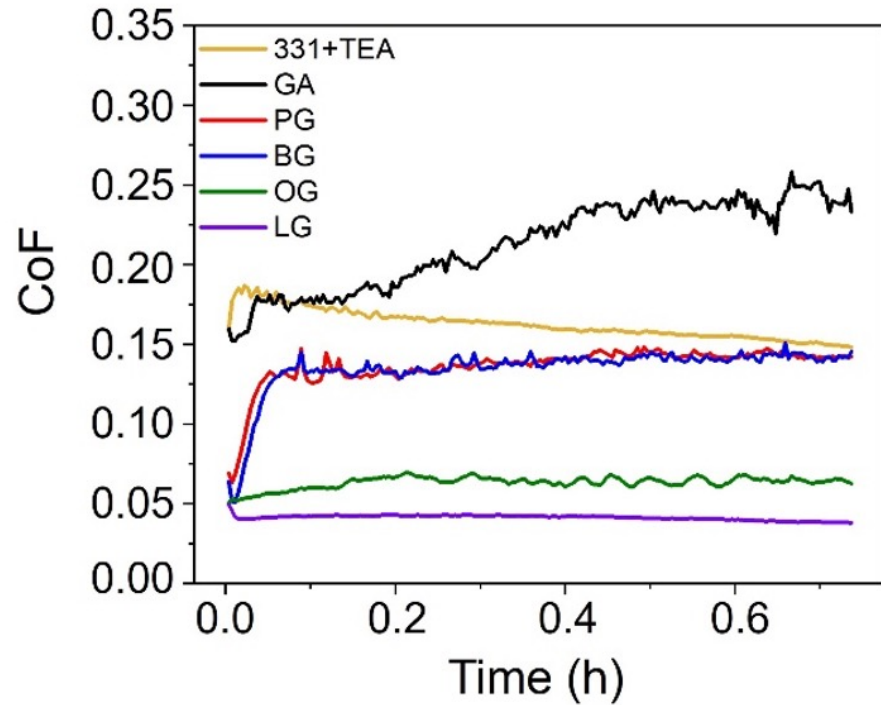
High accuracy needed in  
DFT for training NN  
involving **iron**

A. Pacini, M. Ferrario, S. Loehlè, M.C. Righi,  
Computational Materials Today 1, 100005  
(2024)

1. R = H – gallic acid (GA)
2. R = C<sub>3</sub>H<sub>7</sub> – propyl gallate (PG)
3. R = C<sub>4</sub>H<sub>9</sub> – butyl gallate (BG)
4. R = C<sub>8</sub>H<sub>17</sub> – octyl gallate (OG)
5. R = C<sub>12</sub>H<sub>25</sub> – lauryl gallate (LG)



# Measured friction and wear depend on the chain length

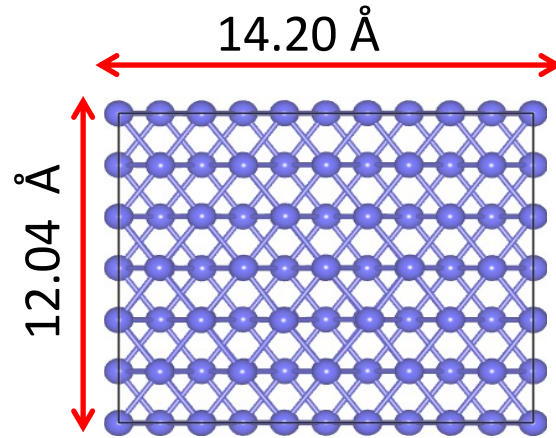


Tribological experiments by M. I. De Barros Bouchet and J. M. Martin and at LTDS

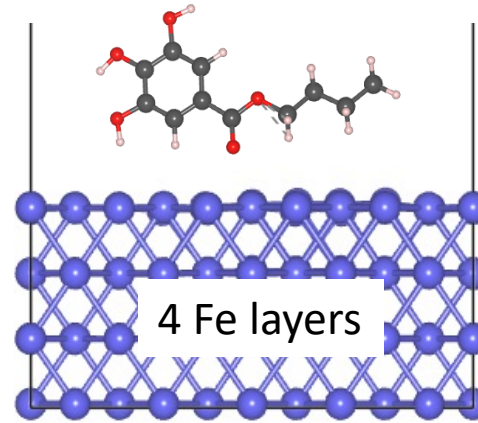
→ The longer the chain, the lower the COF and wear



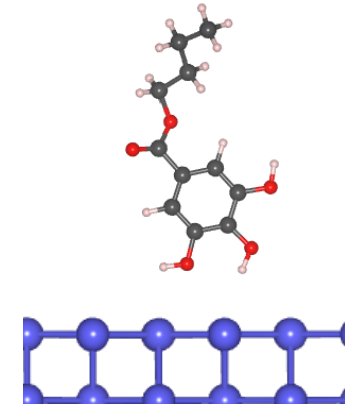
# DFT calculations: Adsorption on Fe(110) surface



Fe(110) surface  
30 atoms per layer,  
Slabs of 4 layers



Parallel orientation  
4 initial structures

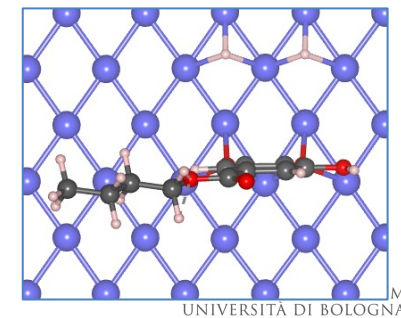
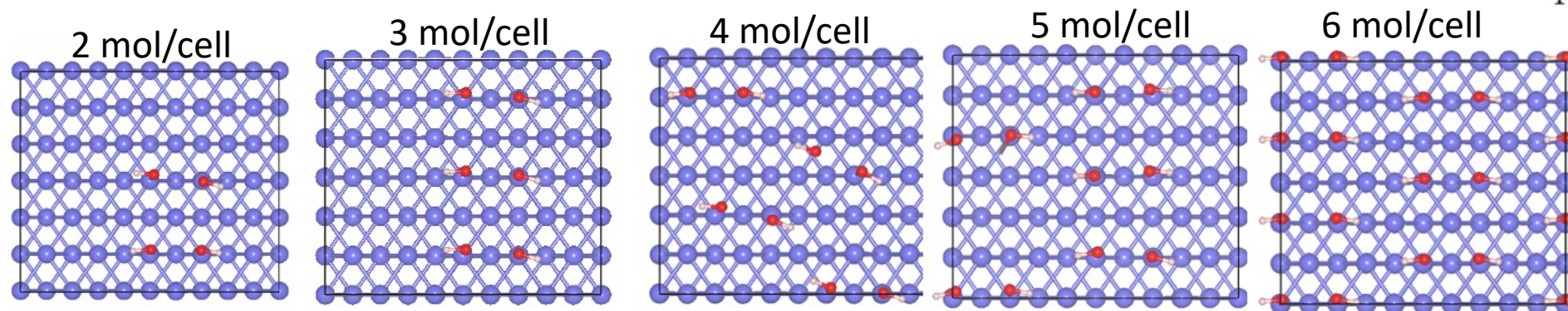
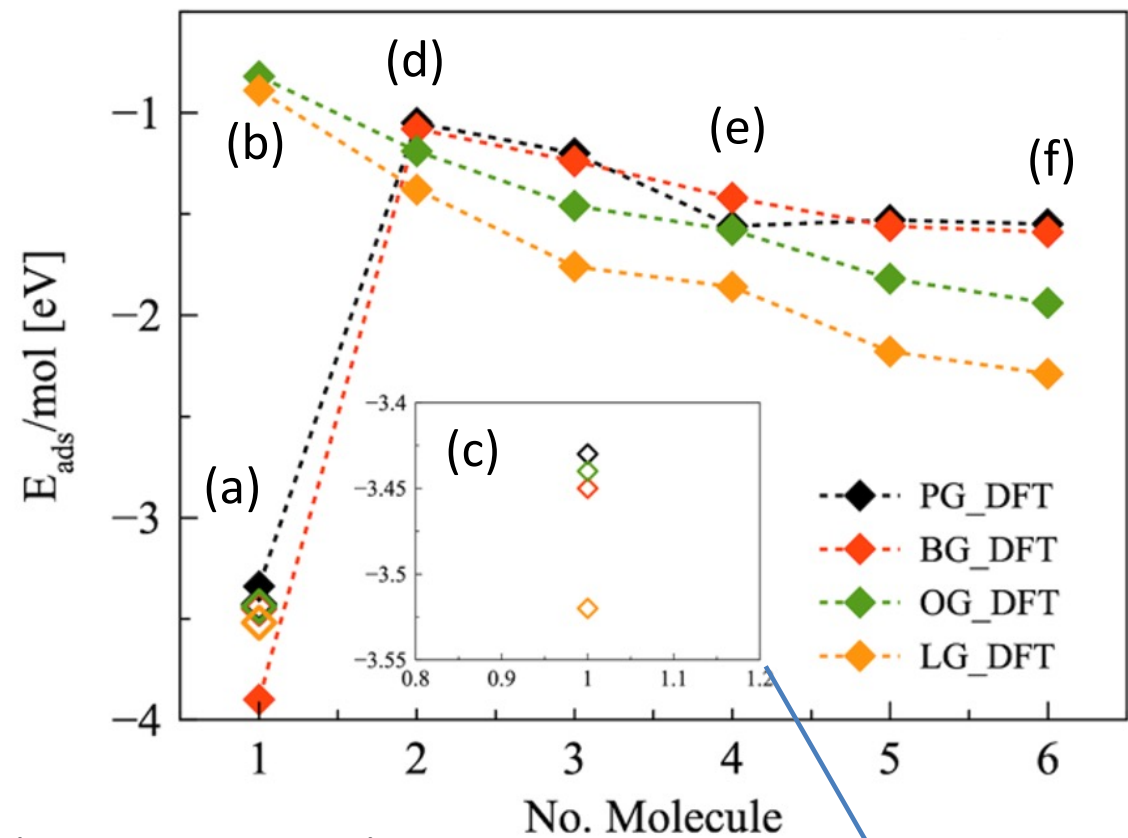
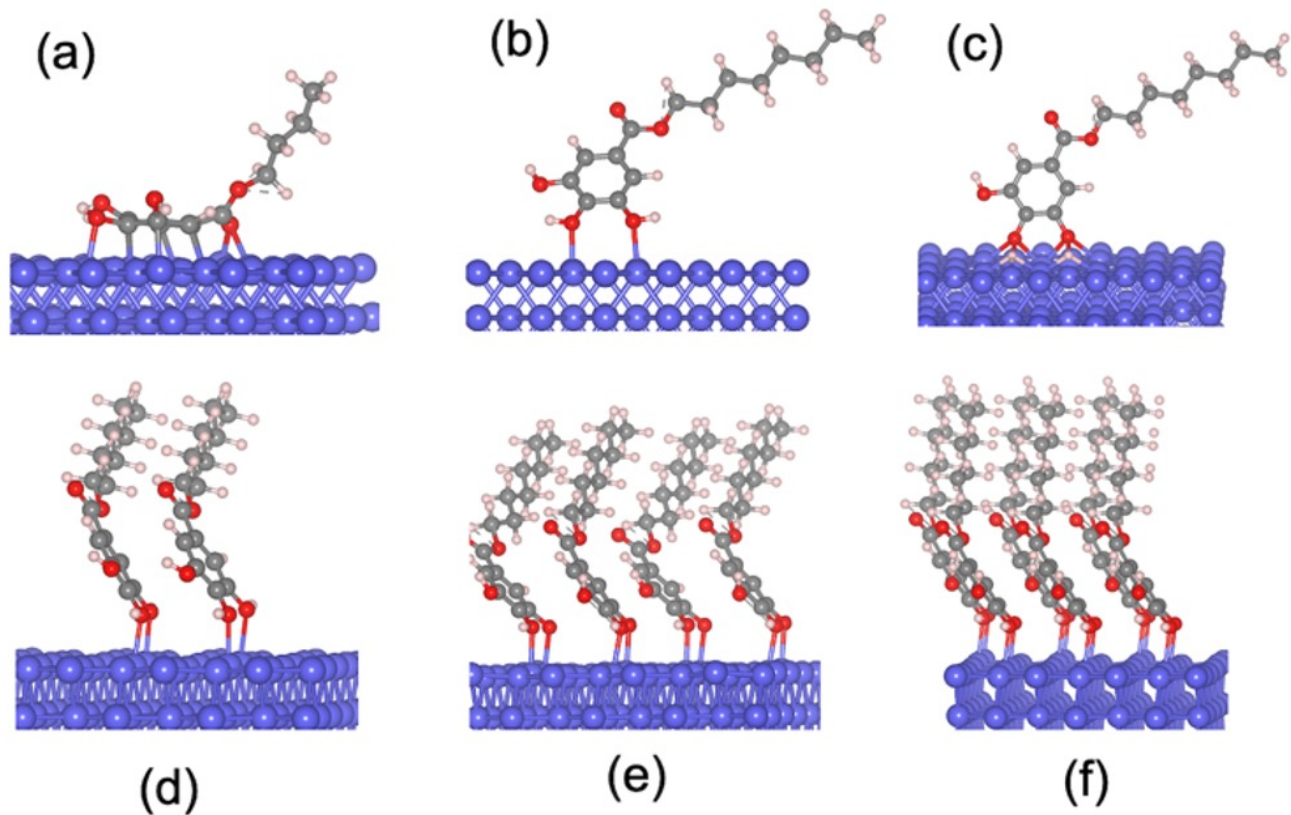


Perpendicular orientation  
4 initial structures

- Gallates adsorption considering different orientations and coverages.
- DFT calculations using VASP package, projector-augmented wave (PAW) method, spin-polarized calculations, cutoff energy to 450 eV, GGA-PBE functional, and vdw-D2 corrections.
- Adsorption energy:  $E_{\text{ads}} = E_{\text{total}} - E_{\text{BG}} - E_{\text{Fe}}$



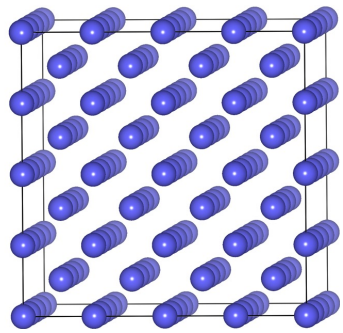
# Results on molecular adsorption



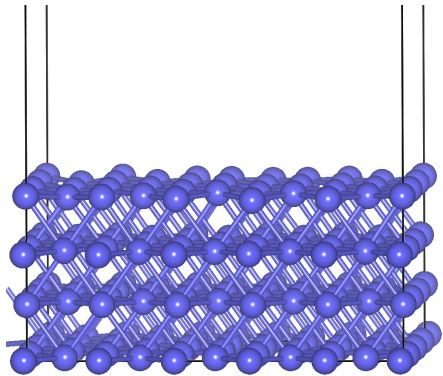


# Data for training the initial NN

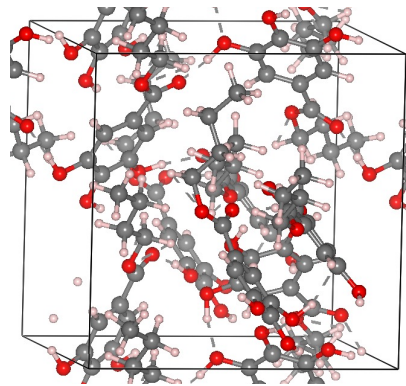
- Initial data: completely from AIMD
- 5 systems with different densities, chemical components, and temperatures
- AIMD simulations at 300, 500, 1000, and 3000 K



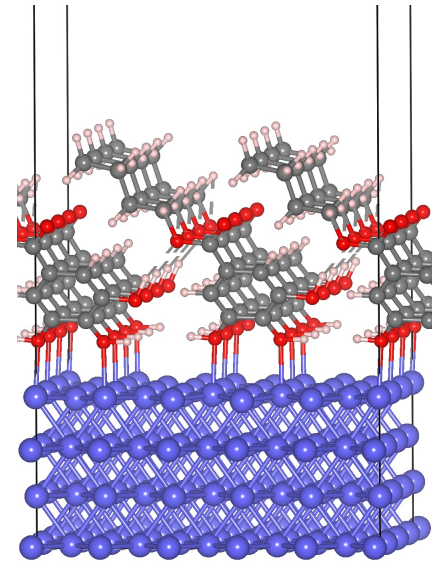
Bulk Fe



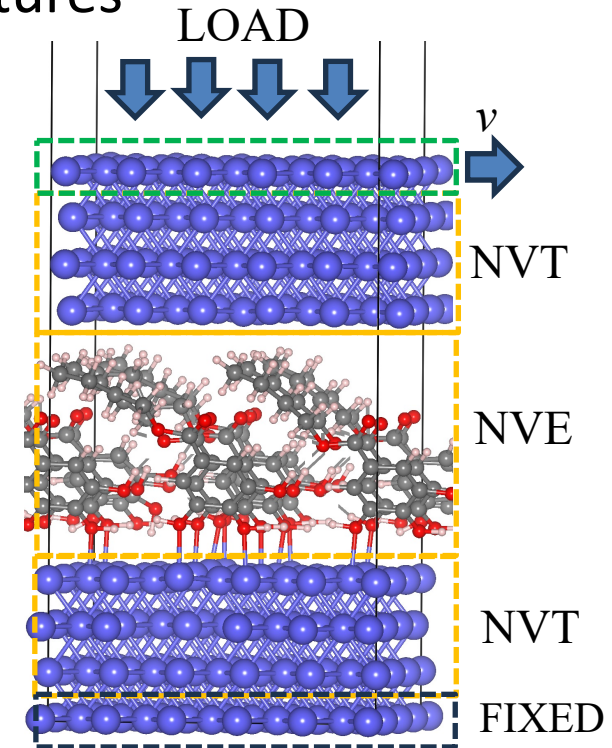
Fe surface



Gallate molecules



Molecules on surface

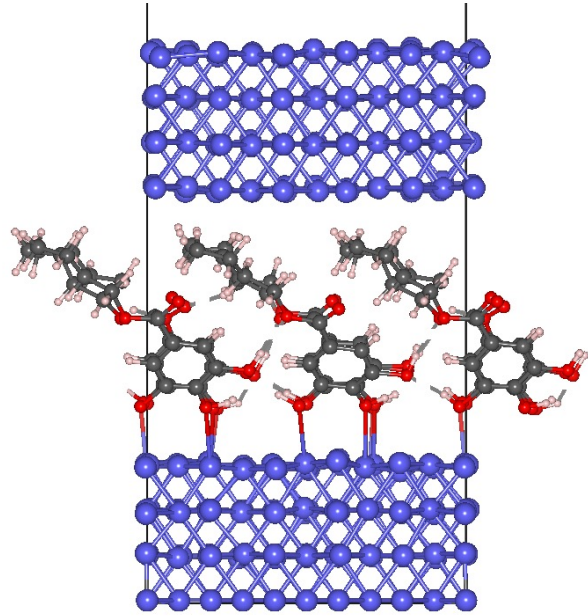


Molecules at the tribo interface



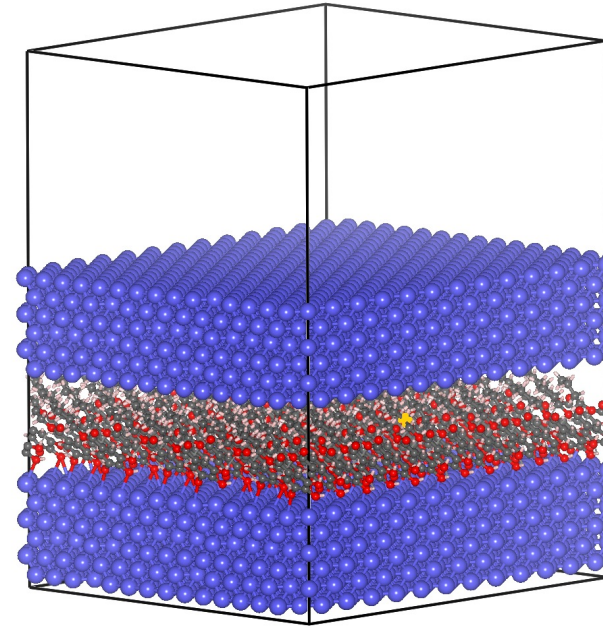
# From DFT to NN and MD simulations

Active learning and Testing



Fe interf. + 6 molecules  
(420 atoms)

MD – production run



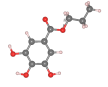
Fe interf. + 72 molecules  
(6480 atoms)

- 0.2 ns equilibration at  $T = 300$  K followed by 1 ns of sliding
- Five different loads applied: 0.5 GPa, 1.0 GPa, 1.5 GPa, 2.0 GPa, 2.5 GPa
- Three different coverages: 100%, 66%, 50%
- Four different molecules: PG, BG, OG, LG

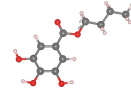
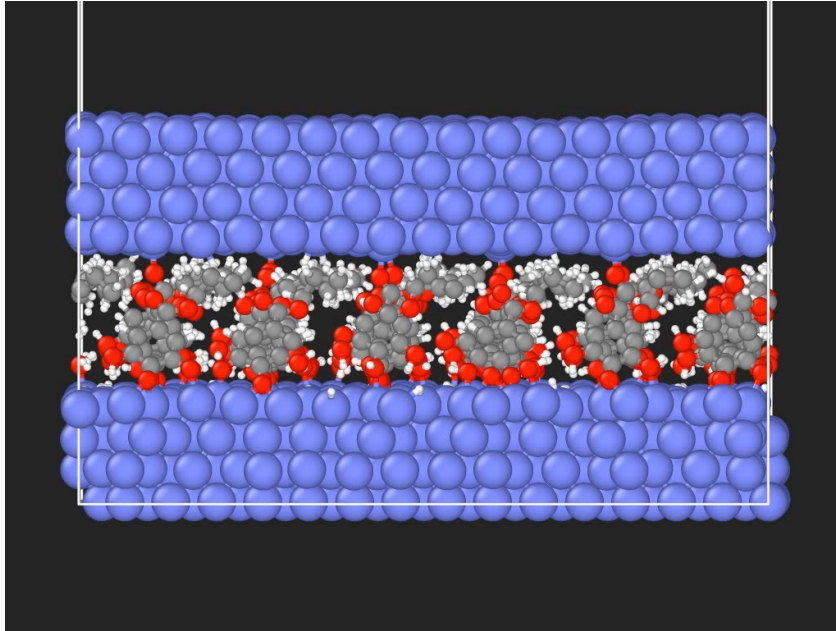
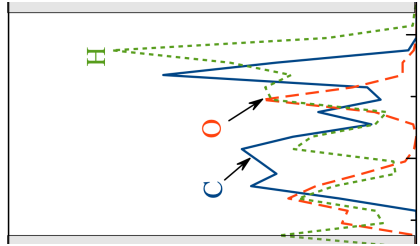




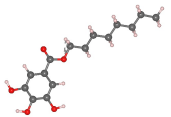
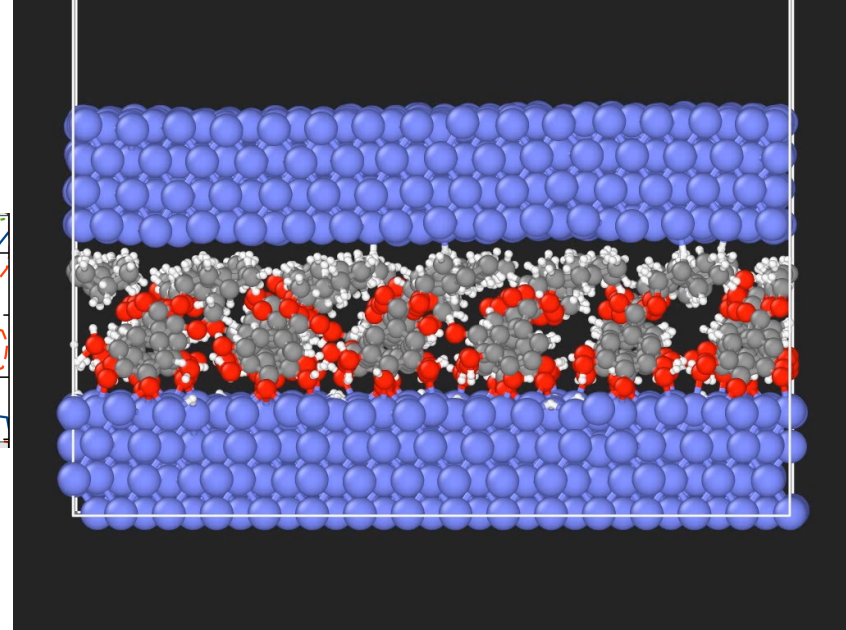
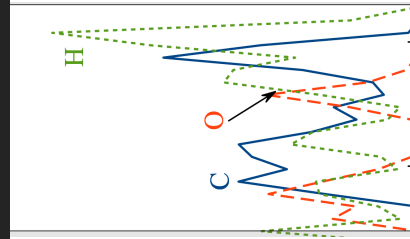
# 100% coverage, 2GPa



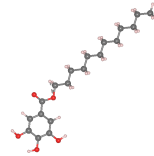
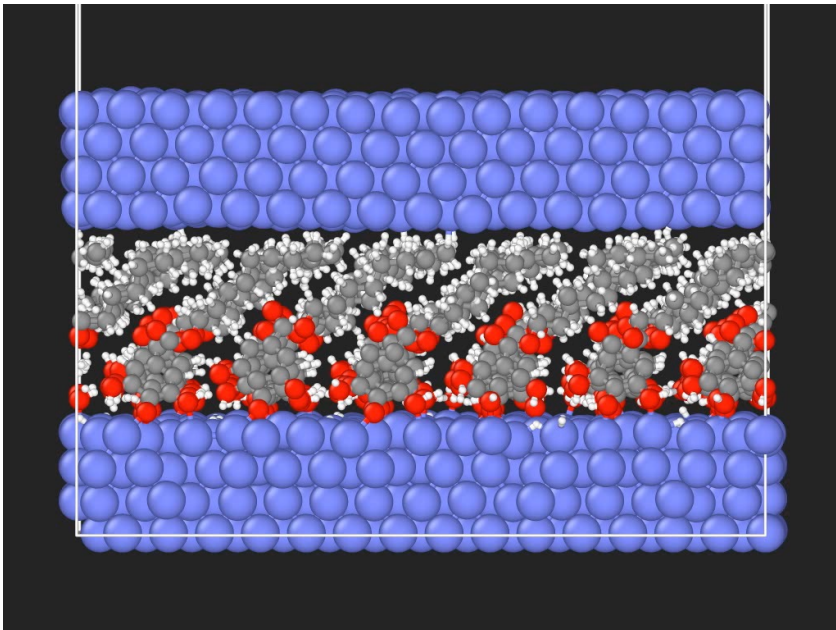
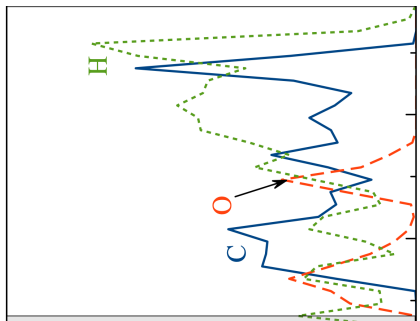
PG



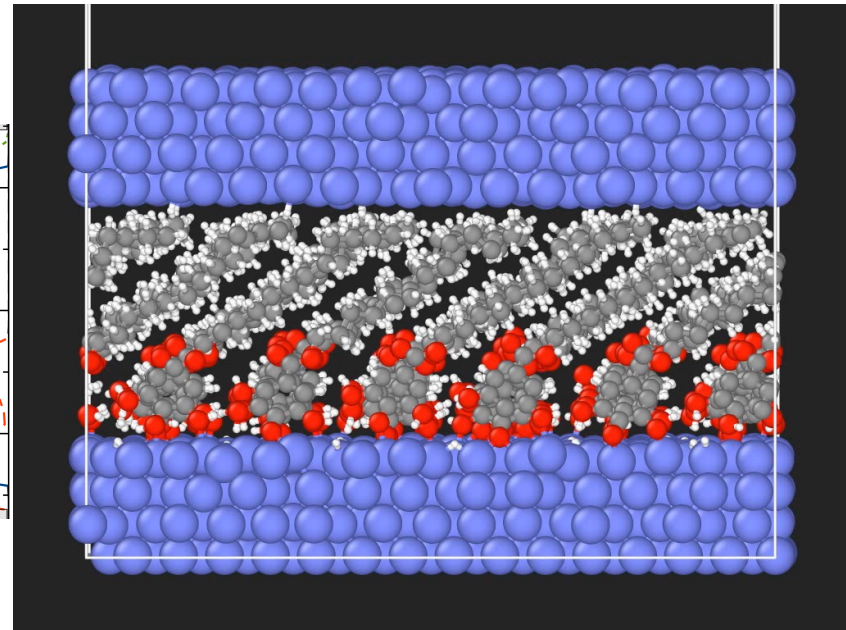
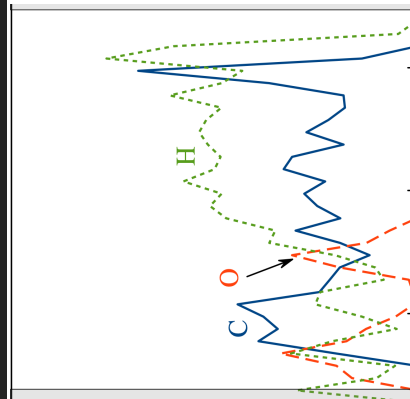
BG



OG

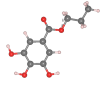


LG

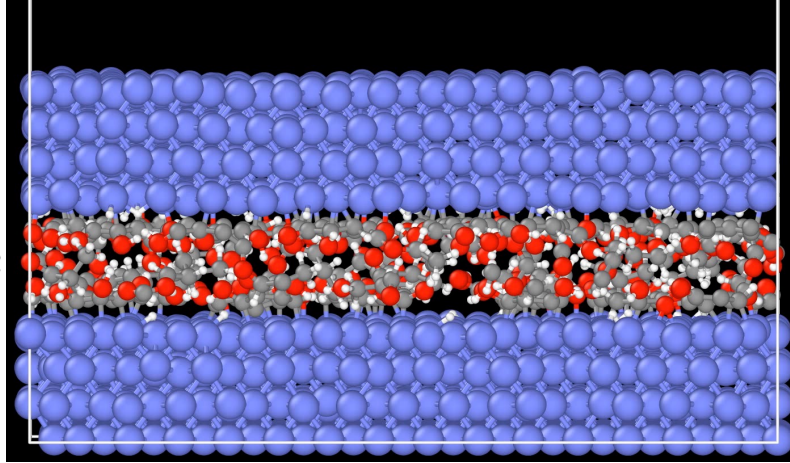




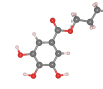
66% coverage



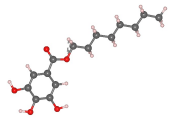
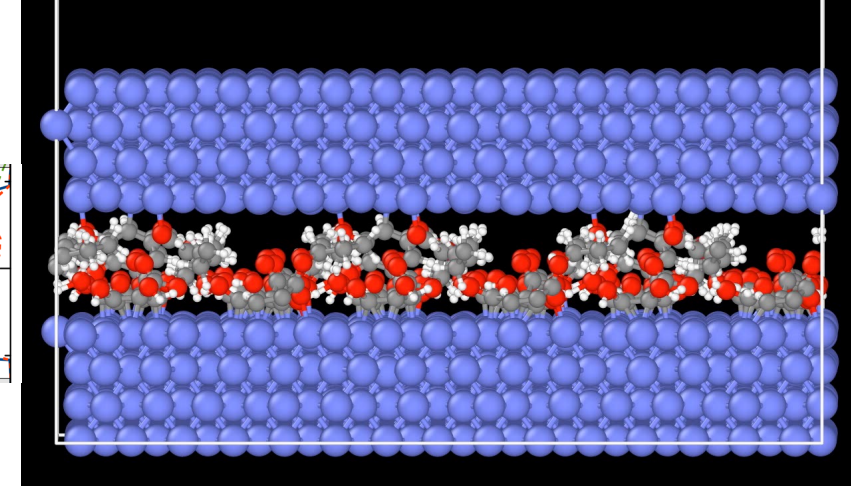
PG



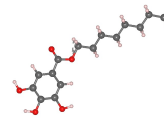
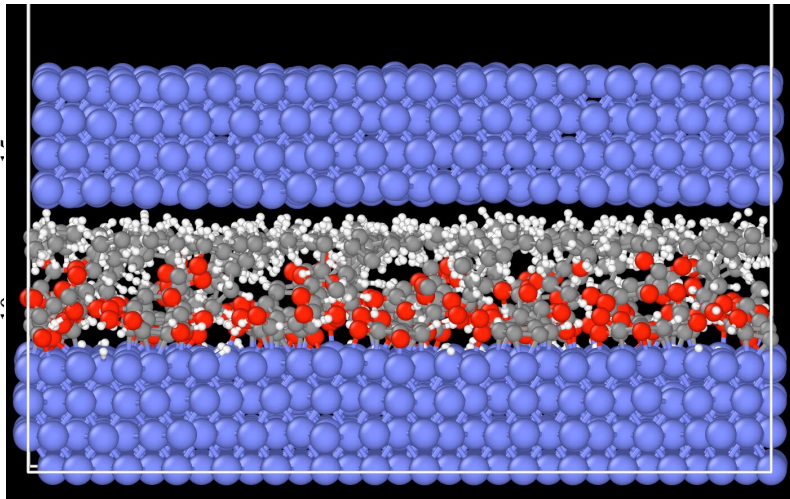
50% coverage



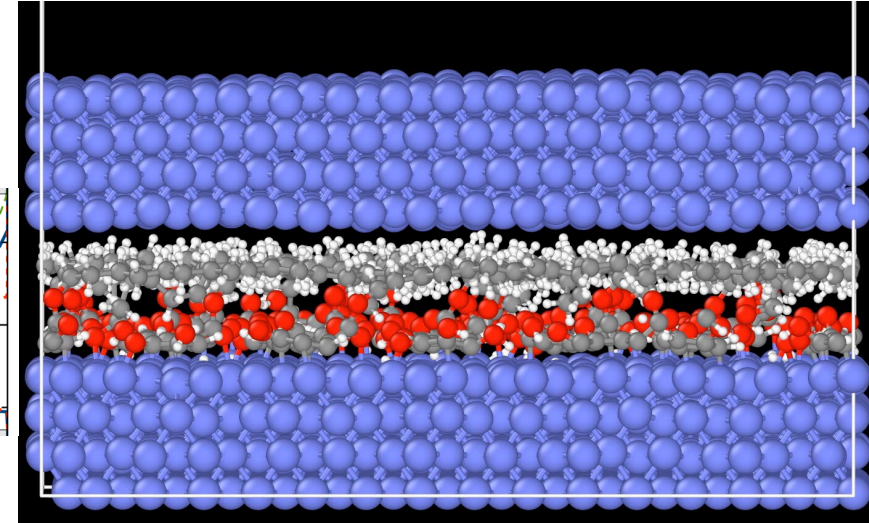
PG



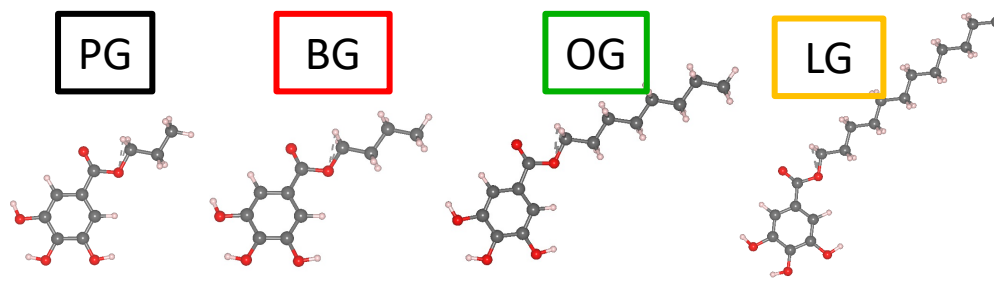
OG



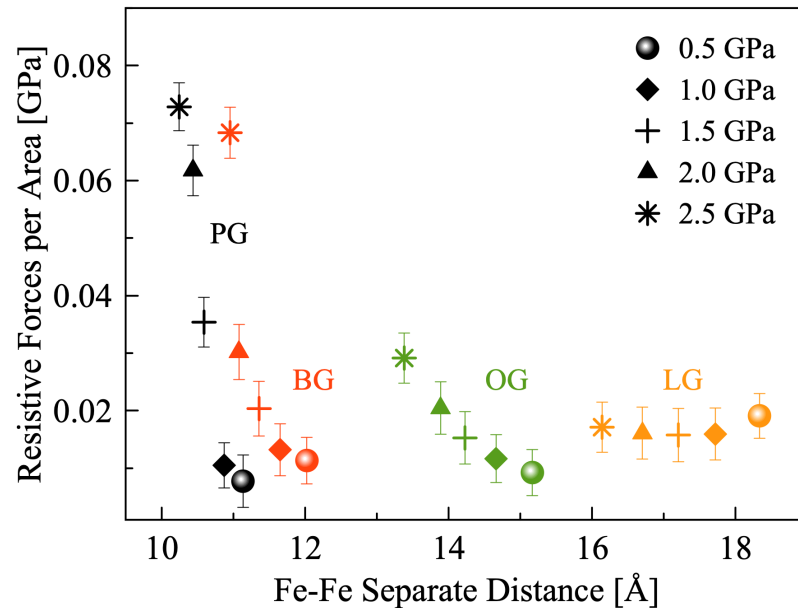
OG



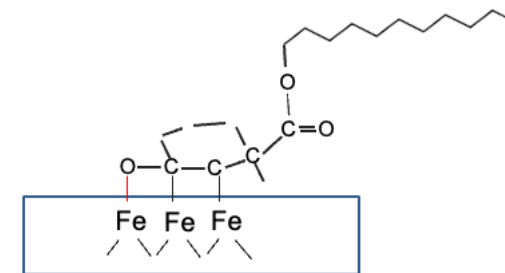
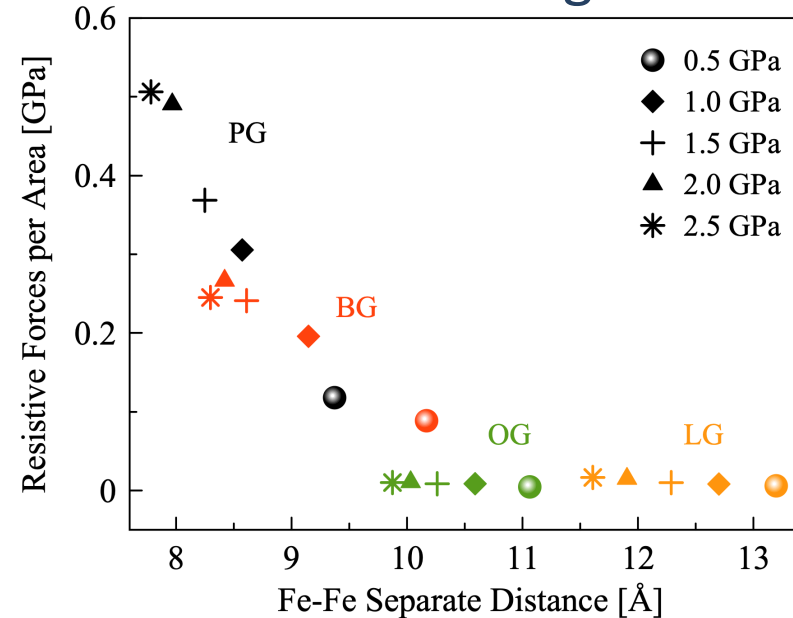
# Effect of chain length explained



100% coverage



66% coverage



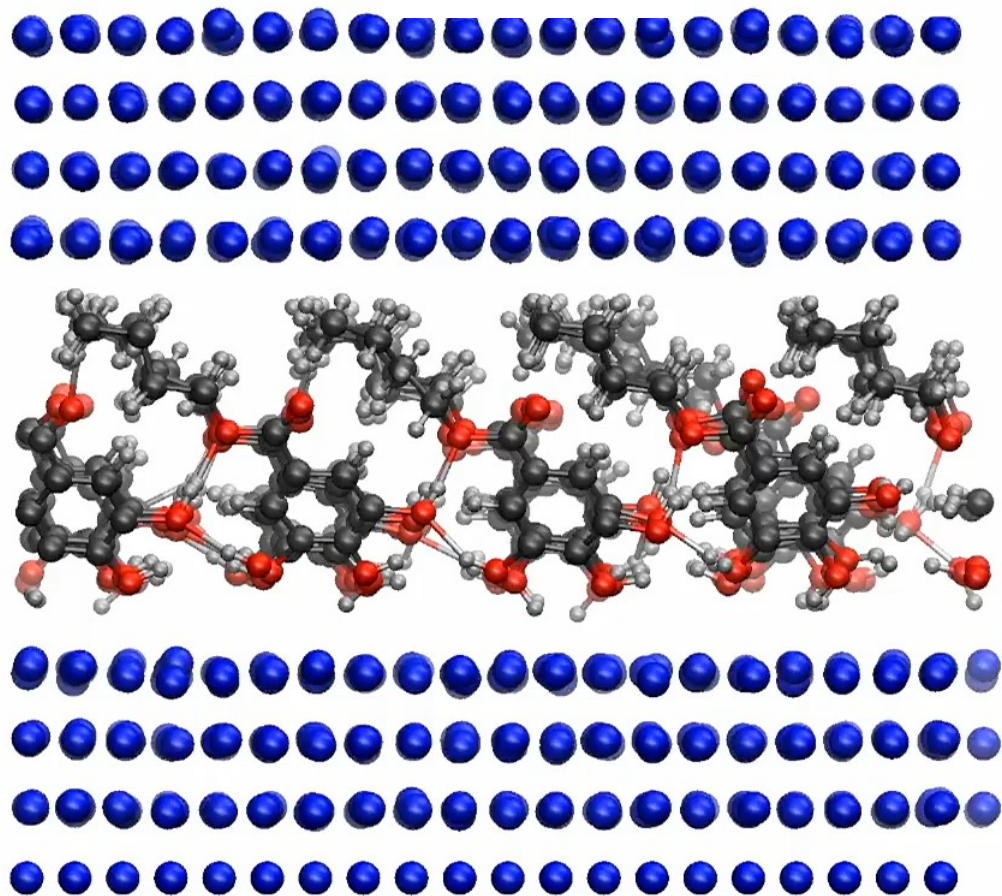
→ The longer the chain, the higher the interfacial separation, the lower the friction force.



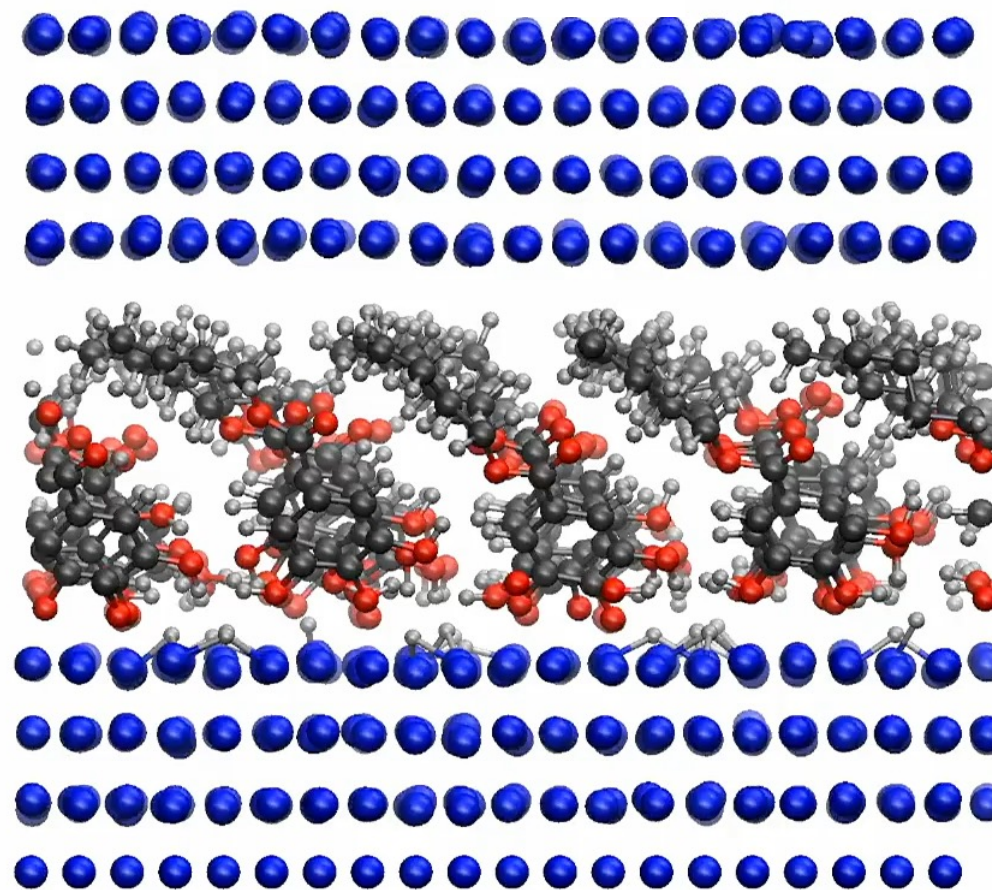
# Comparison with ReaxFF

28BG at 0.5 GPa, 300 K

## ReaxFF



## NN

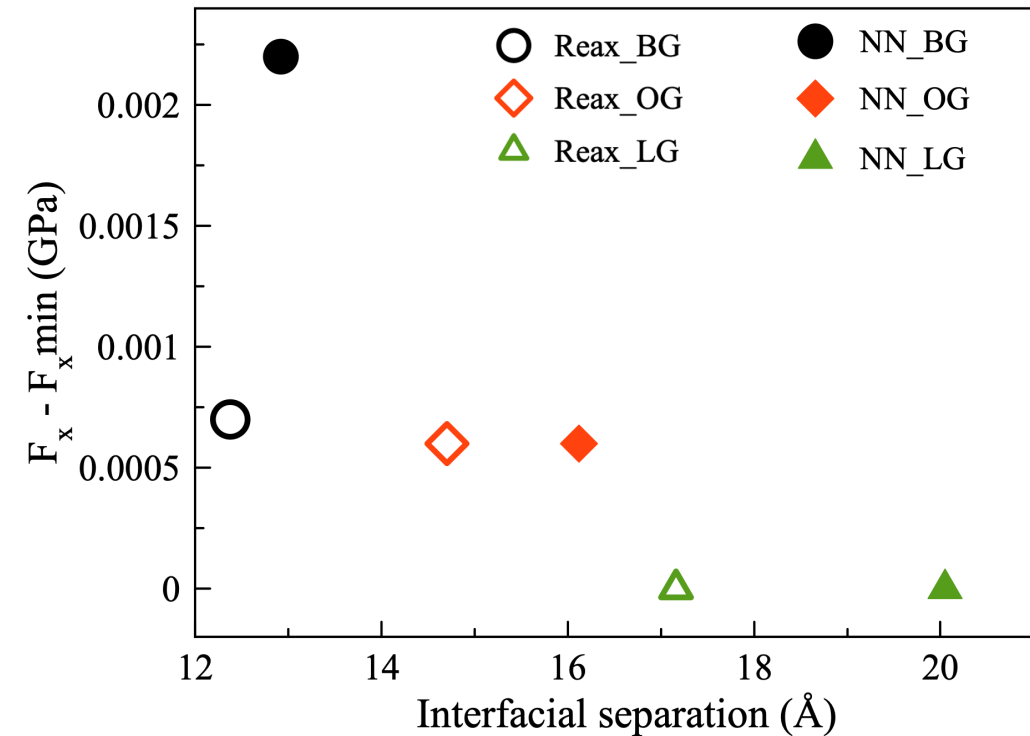
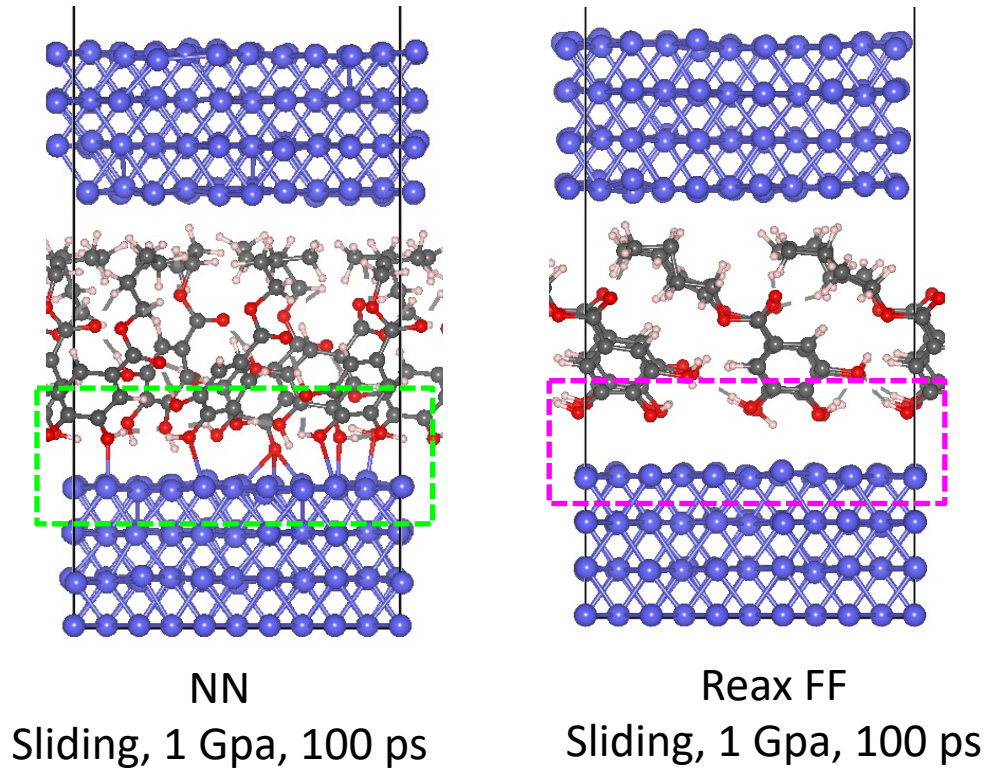


ReaxFF potential:

- Y.K. Shin, H. Kwak, A.V. Vasenkov, D. Sengupta and A.C.T. van Duin *ACS Catalysis*, 2015, 5 (12), pp 7226-7236
- (Fe parameters identical to M. Aryanpour, van Duin and Kubicki, *J Phys Chem A*, 2015, 114, 6298-6307 )
- [QEQ charge equilibration (Rappe & Goddard) following Aktulga, Fogarty, Pandit, Grama, *Parallel Computing*, 2012, 38, 245-259]



# The effect of chain length not captured by ReaxFF



- Reaxff does not allow H detachment and molecular docking to the substrate
- Relative motion of the molecules to the substrate
- No effect of chain length



# Ab initio studies on diamond– silica interfaces

## 1. Tribochemistry of silica-diamond interface

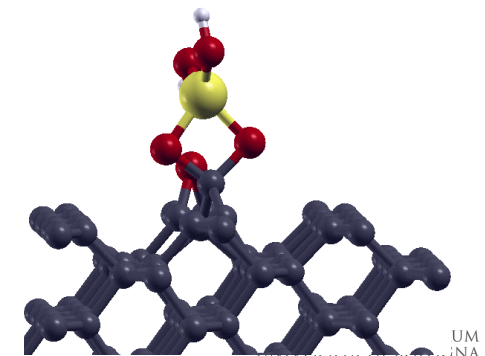
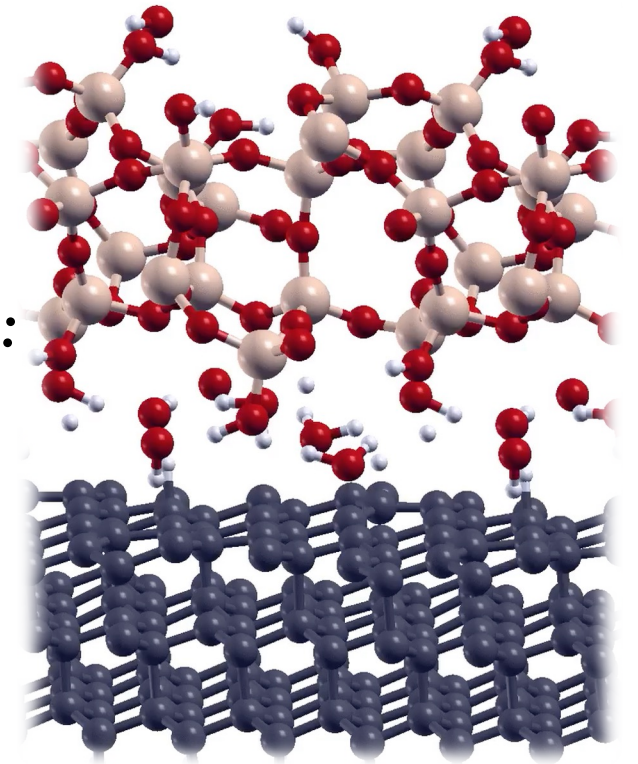
- *Adhesion, Friction and Tribochemical reactions at the Diamond-Silica Interface*, *Carbon* **203**, 601 (2023)
- *Nanotribological Properties of Oxidized Diamond/Silica Interfaces: Insights into the Atomistic Mechanisms of Wear and Friction by Ab Initio Molecular Dynamics Simulations*, *ACS Applied Nano Materials* **6**, 16674 (2023)

## 2. Atomistic mechanisms of diamond wear

- *Atomistic Wear Mechanisms in Diamond: Effects of Surface Orientation, Stress, and Interaction with Adsorbed Molecules*, *Langmuir* **39**, 14396 (2023)

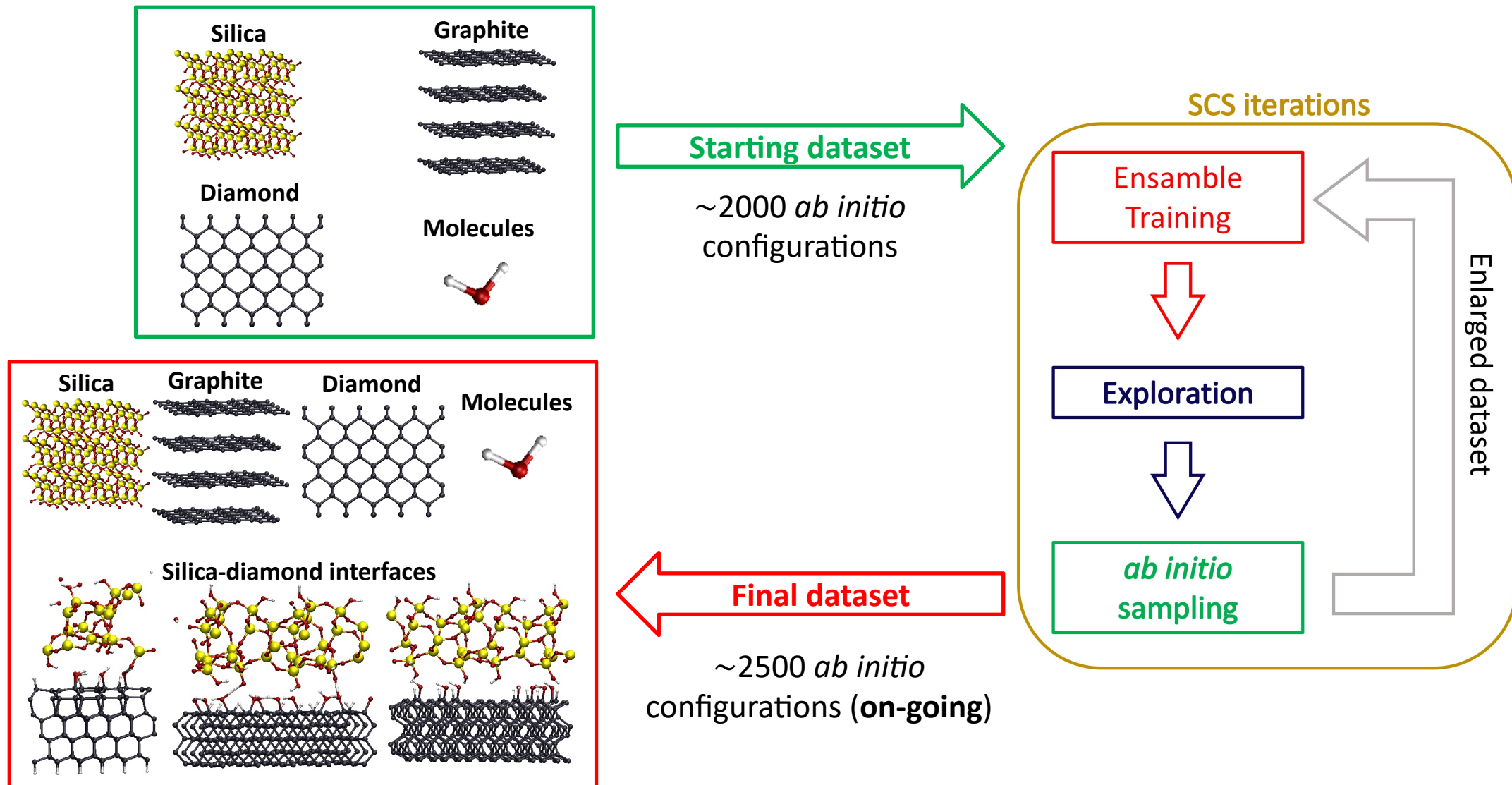
## 3. Adsorption and dissociation of H<sub>2</sub>, H<sub>2</sub>O and O<sub>2</sub> on diamond

- *Ab initio insights into the interaction mechanisms between H<sub>2</sub>, H<sub>2</sub>O, and O<sub>2</sub> molecules with diamond surfaces*, *Carbon* **199**, 497 (2022)
- *Tuning the adsorption of H<sub>2</sub>O, H<sub>2</sub> and O<sub>2</sub> molecules on diamond surfaces by B-doping*, *Surfaces and Interfaces* **46**, 104105 (2024)



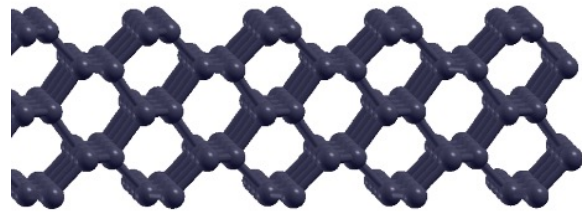
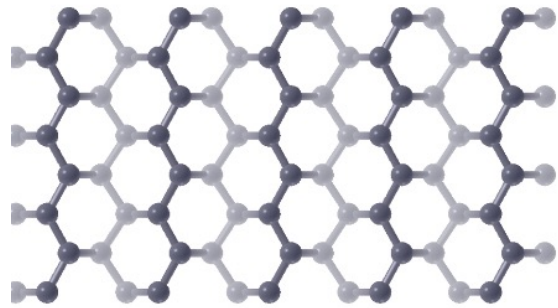
# Active learning with SCS (ongoing)

We use our in-house developed workflow, *Smart Configuration Sampling (SCS)* to perform active learning



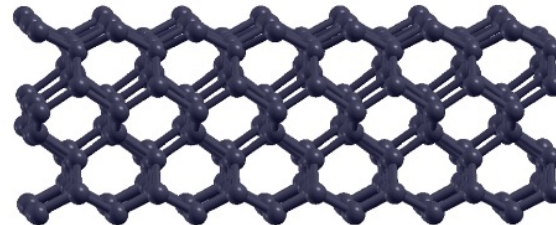
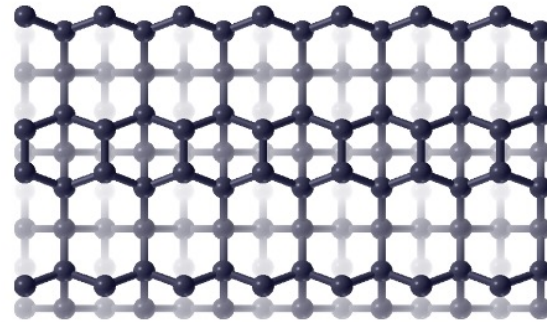
# Effects of diamond orientation (on-going)

**C(110)**



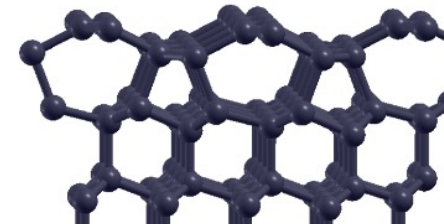
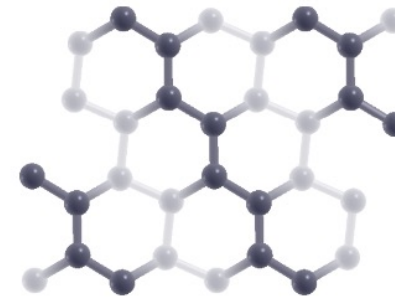
**5.9 J/m<sup>2</sup>**

**C(001)**



**5.7 J/m<sup>2</sup>**

**R-C(111)**



**4.1 J/m<sup>2</sup>**

**Calculated surface energies**

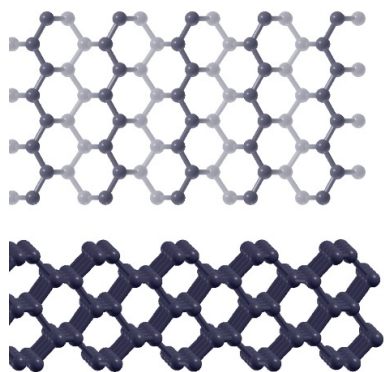




# ML-MD: effects of diamond orientation (on-going)

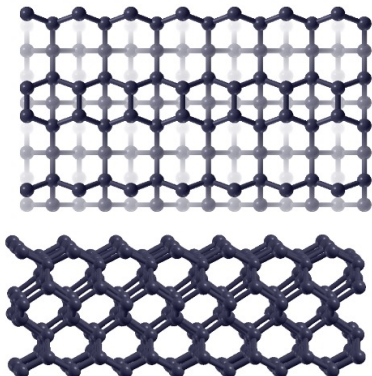
4800 atoms, surf area 0.25 nm<sup>2</sup>  
Simulated time 1 ns  
T= 300K, v= 50 m/s, Load 1 GPa

C(110)



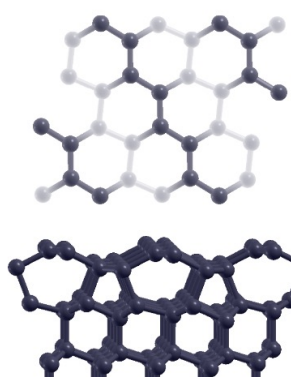
5.9 J/m<sup>2</sup>

C(001)



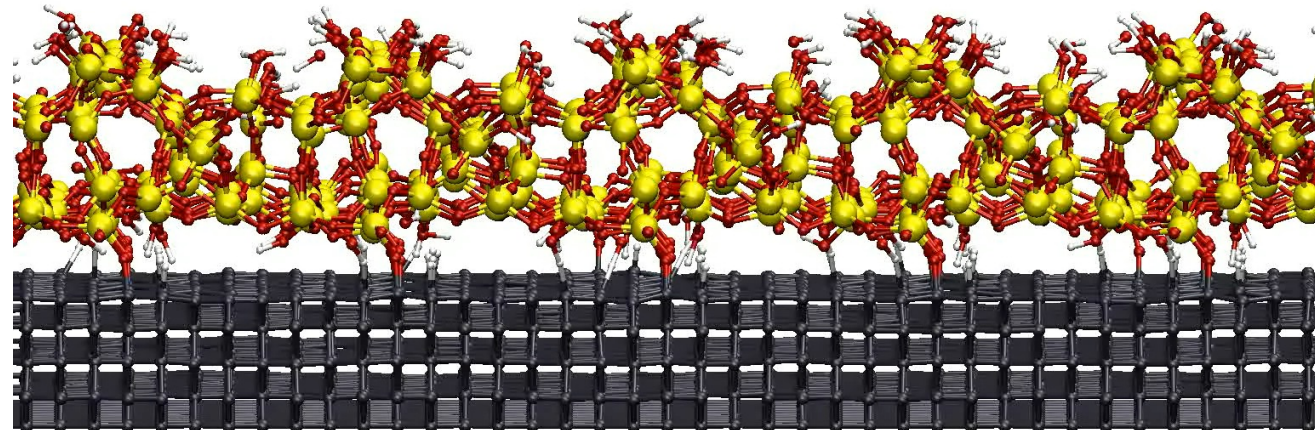
5.7 J/m<sup>2</sup>

R-C(111)

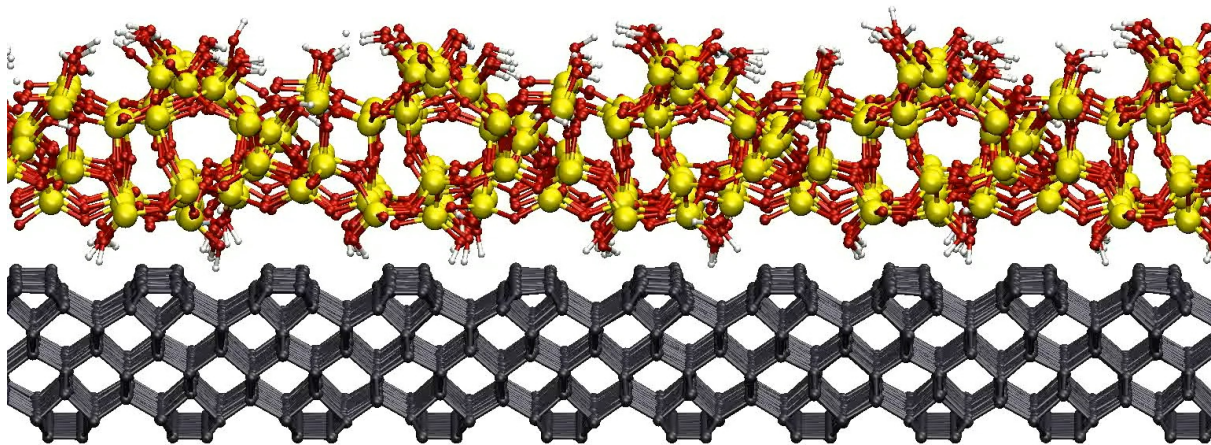


4.1 J/m<sup>2</sup>

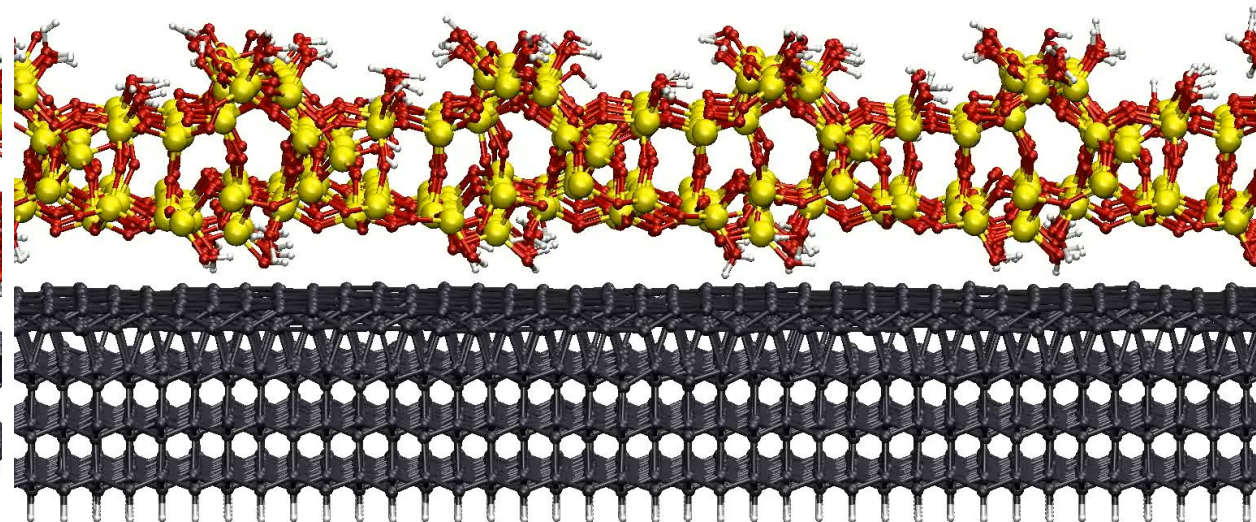
C(110)



C(001)

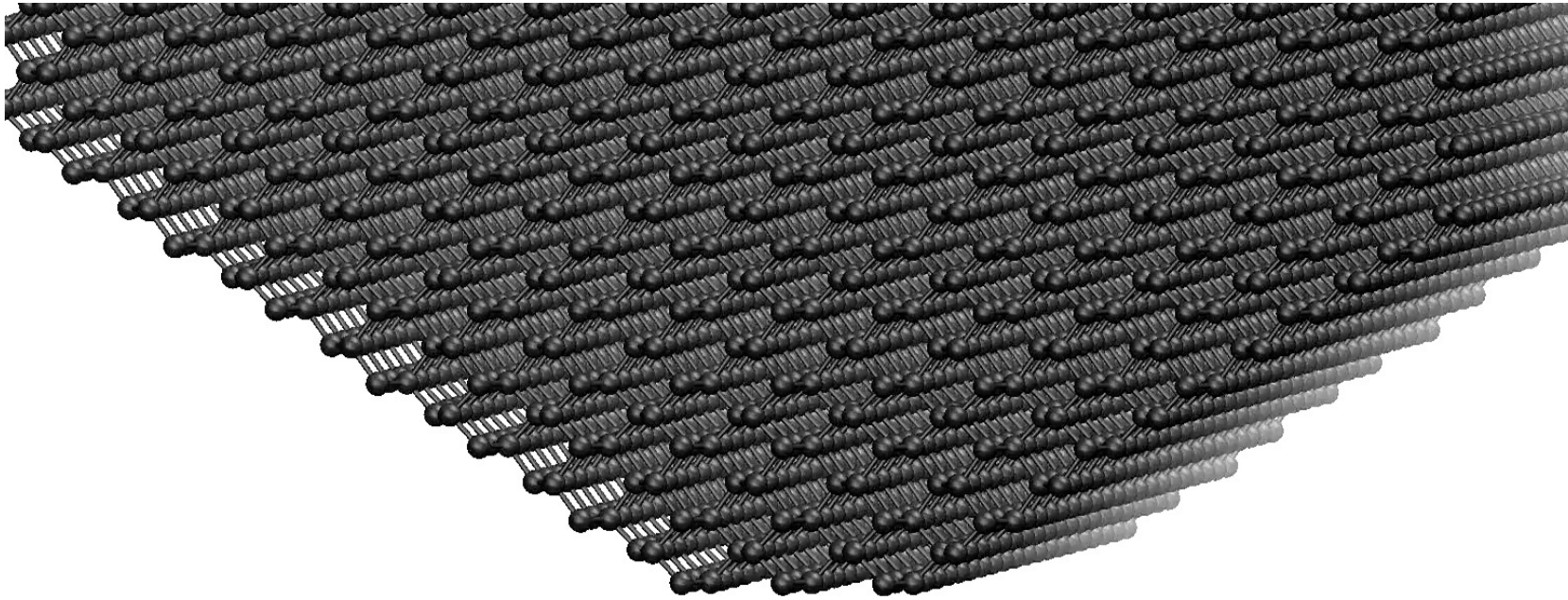


R-C(111)





## ML-MD of a sliding asperity (on-going)

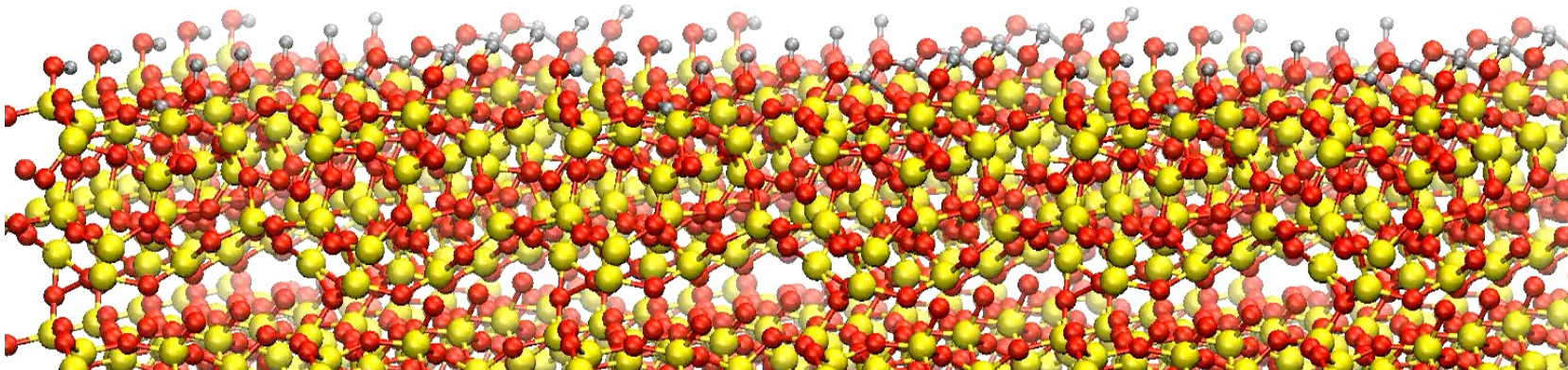


**10850 atoms**

**Simulated time 0, 5 ns/day**

Tip surface C(110)

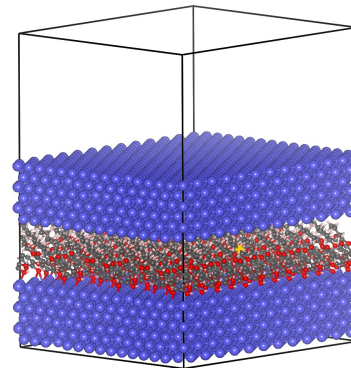
T= 300K, v= 1 m/s, Load 400 MPa



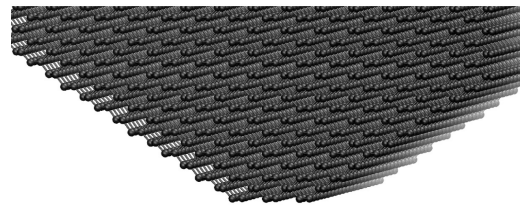
# Conclusions

- Accurate Interatomic Potentials can be obtained by an Active Learning approach.
- We developed software, **SCS**, which couples DeeP-MD, LAMMPS and Quantum Espresso for the active learning training of Neural Networks
- By means of ML-MD we were able to simulate key tribological systems, which are impossible to simulate *ab initio*

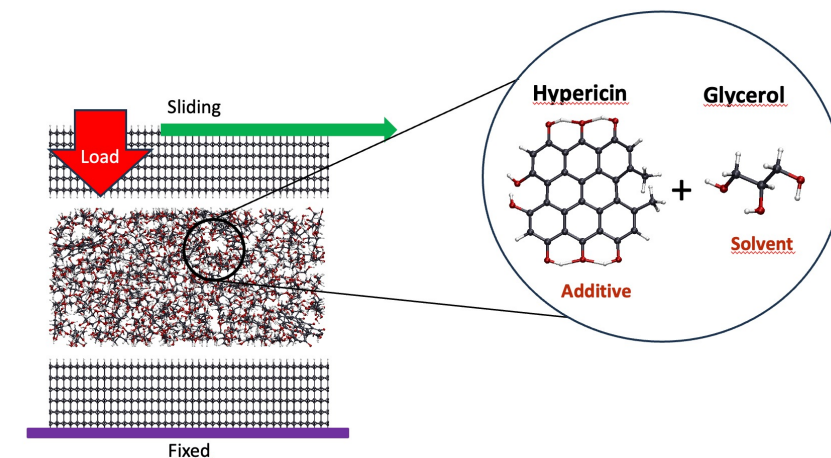
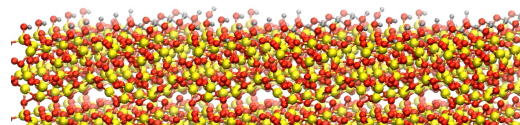
- Self assembled monolayers of large molecules



- Additives included in liquid media



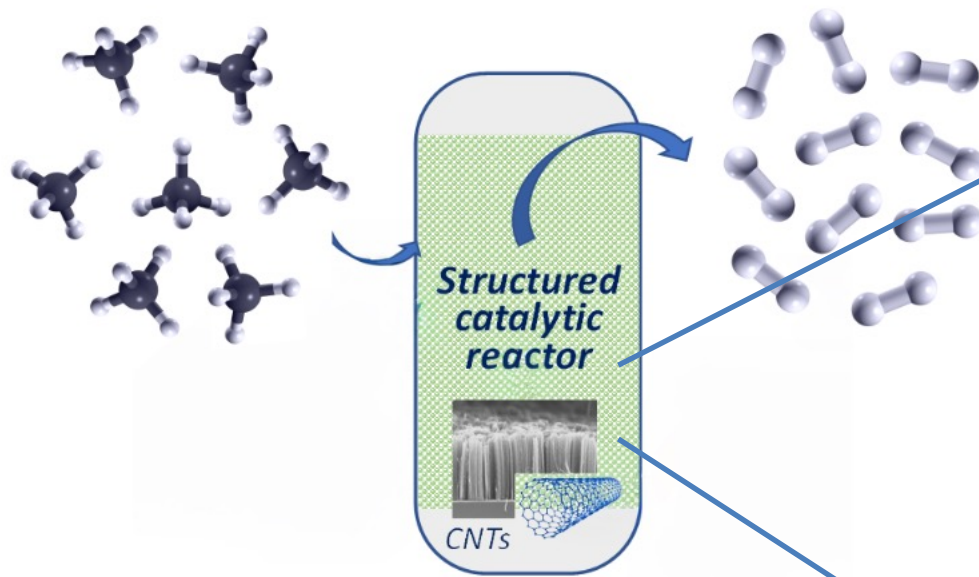
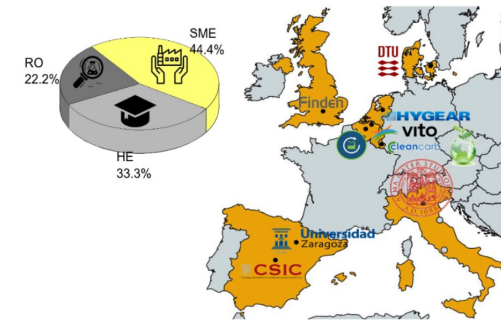
- *In silico* AFM experiments



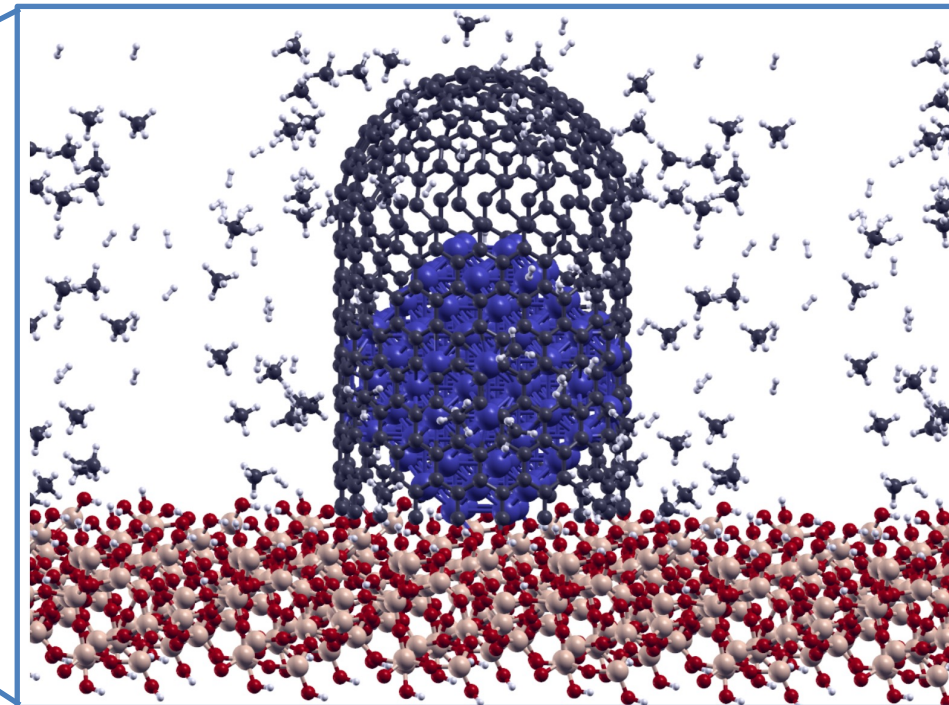


# Simulating what happens in a reactor

EU Project "STructured unconventional reactors for CO2-free Methane catalytic cracking" (**STORMING**)



EUROHPC-JU 20 M core hours grant on Leonardo





ADVANCING SOLID INTERFACES AND LUBRICANTS  
BY FIRST PRINCIPLES MATERIALS DESIGN



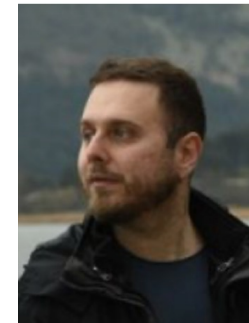
Paolo Restuccia - Edoardo Marquis - Vito Foderà - Francesca Benini -  
Emiliano Poli - Enrico Pedretti - Elisa Damiani - Clelia Righi -  
Margherita Marsili - Alberto Pacini



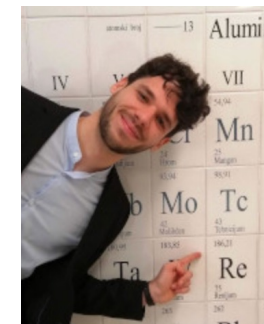
Mauro Ferrario



Huong Ta Thi Thuy



Stefanos Giaremìs



Matteo Vezzelli







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